



# Atmospheric Chemistry of Solar and Extrasolar Gas Giants

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with thanks to the Exoclimatology Theory Group,  
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# Outline

- Big questions
- Some chemical detection firsts with JWST
- Sulphur dioxide ( $\text{SO}_2$ )
  - What we can learn from  $\text{SO}_2$
- Methane ( $\text{CH}_4$ )
  - The “missing methane” problem
- Chemical networks
- Lessons from the Solar System

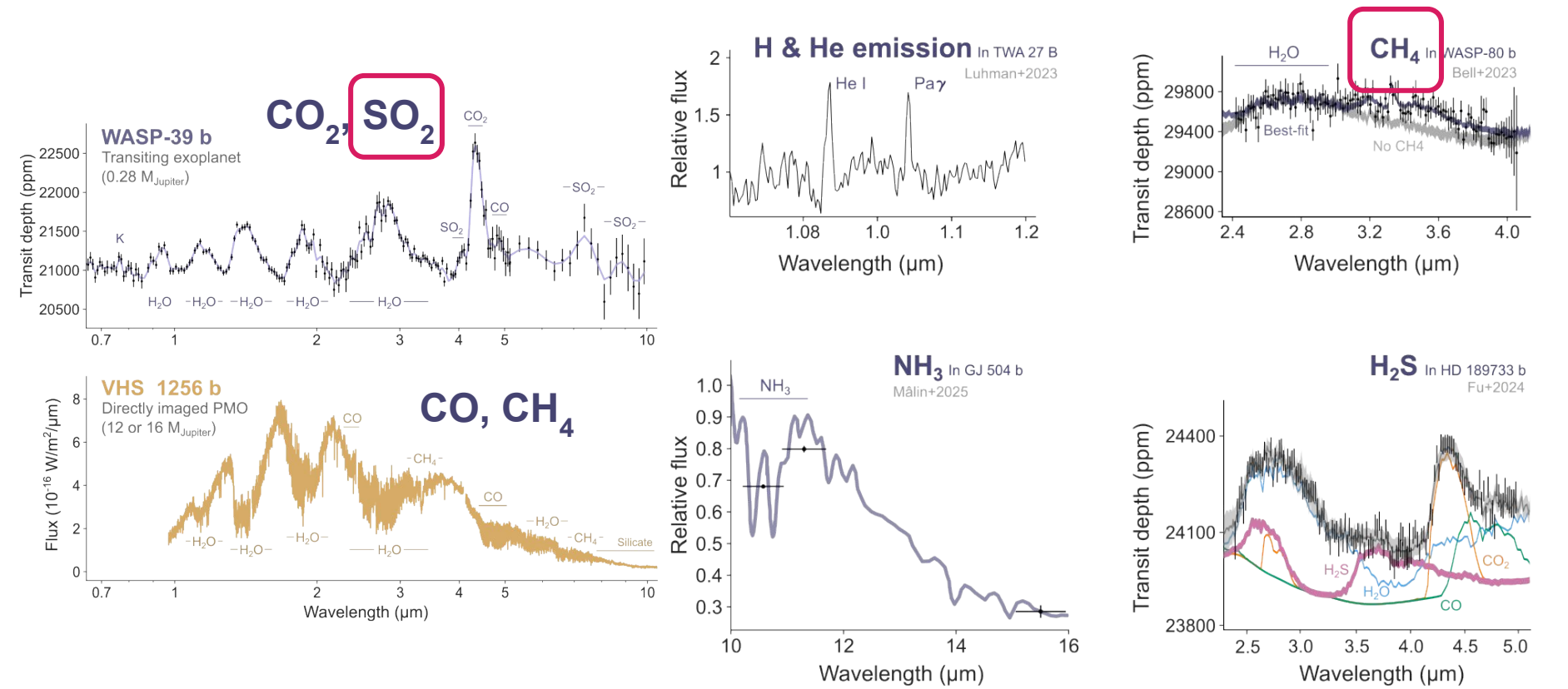


# Big questions

- How do planets form?
- How do planets get their atmospheres?
- How chemically diverse are planetary atmospheres?
- What processes affect atmospheric composition?

**What can the observed atmospheric composition tell us about the past and the present of a planet?**

# JWST extrasolar-gas-giant firsts



# Why do we care about **sulphur**?

S is abundant and of intermediate volatility

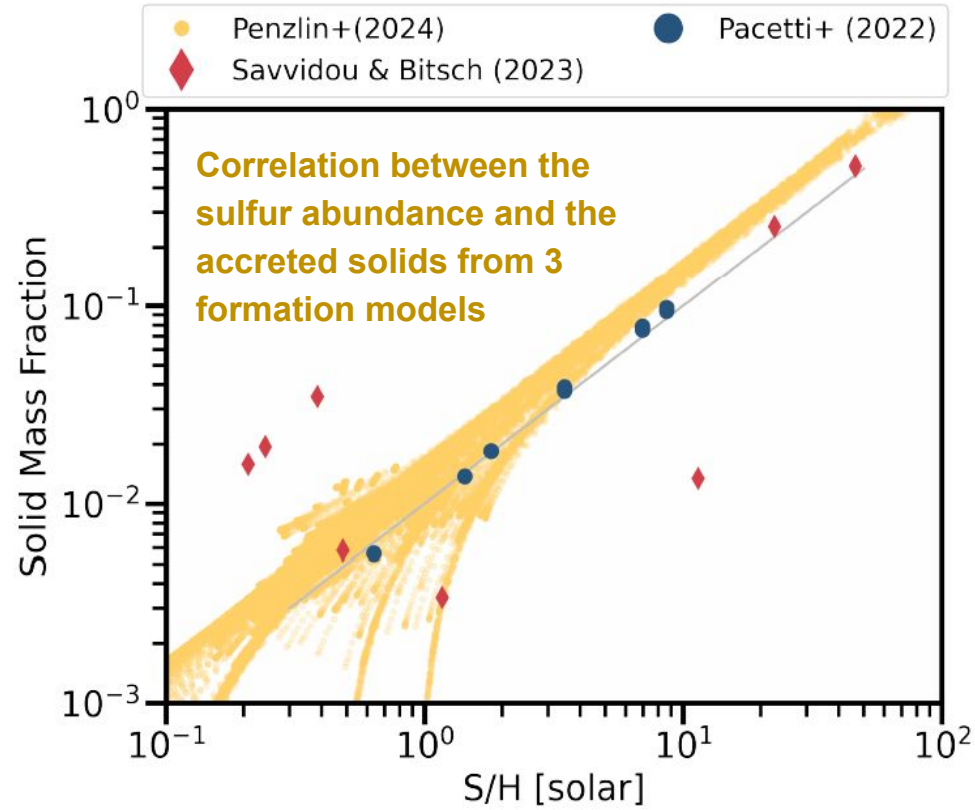
- tracer for the **amount of solid material accreted by planets** during their formation
- helps **distinguish between competing models of planet formation**, complementing C, O, Si, Na, Fe

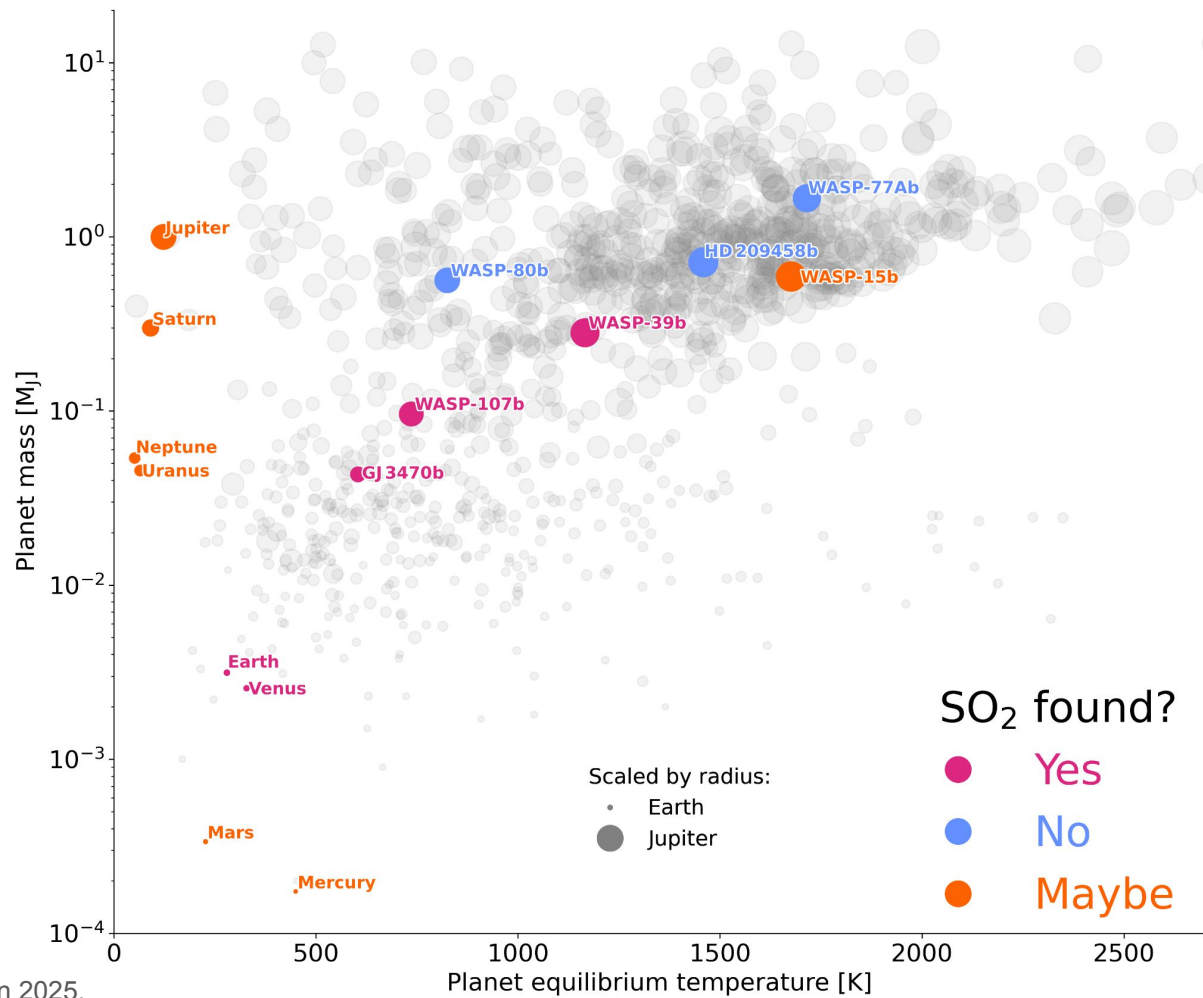
Planet formation problem: **S can sublime into  $\text{H}_2\text{S}$**  at lower temperatures and may **modify planet's S/O**

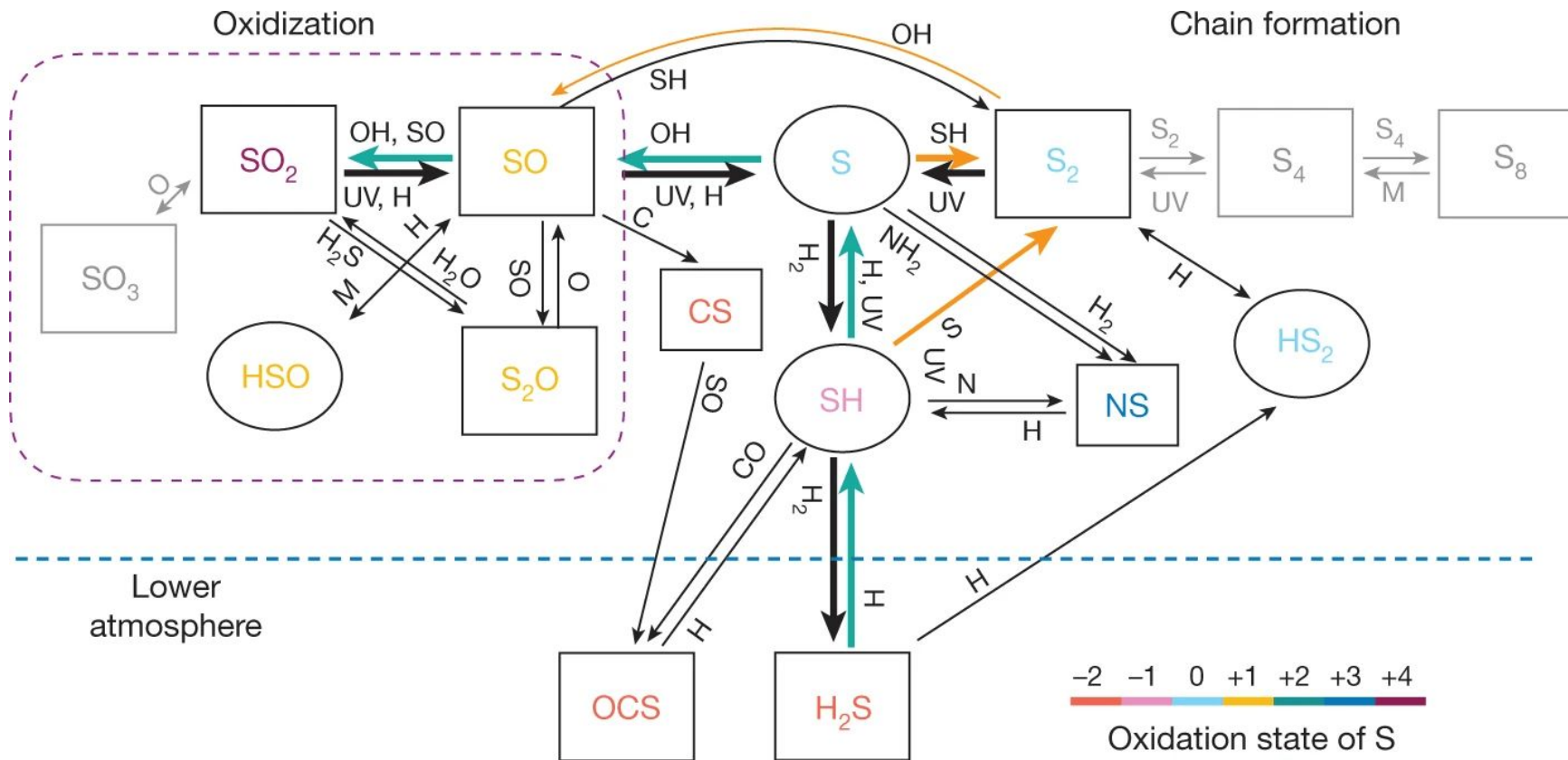
S chemistry in  $\text{H}_2$ -rich atmospheres is **poorly known**

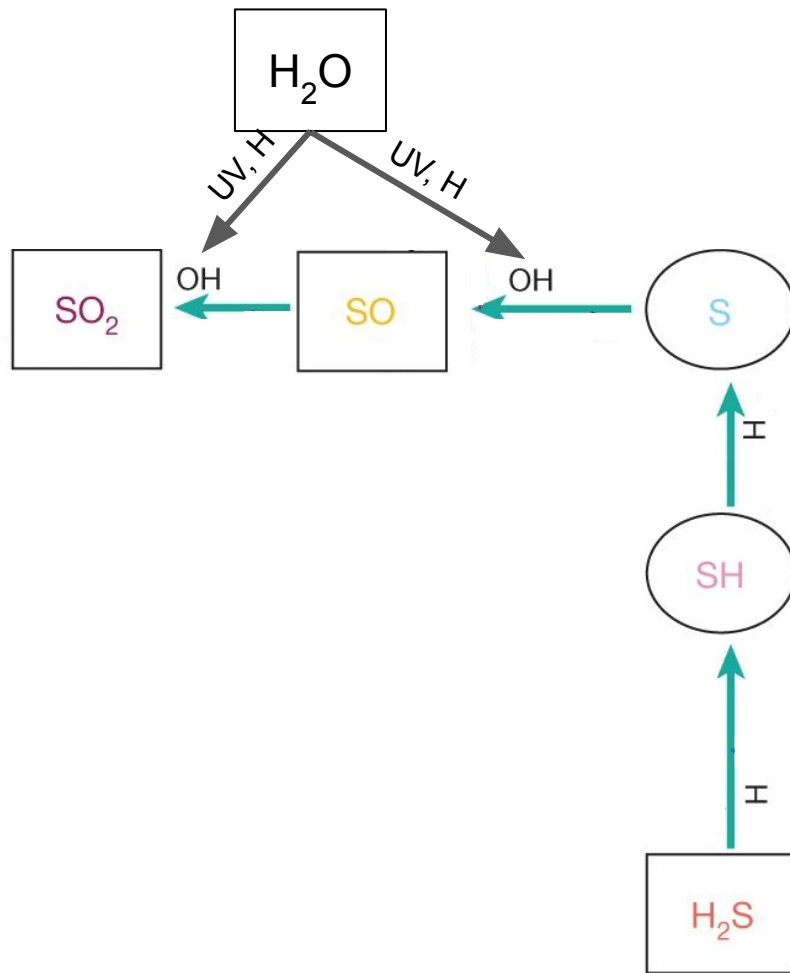
- **$\text{SO}_2$**  is a **thermodynamically unstable** form of sulfur, because it is highly oxidised
- **$\text{SO}_2$**  detection means that  $\text{H}_2\text{S}$ , a thermodynamically stable form of sulfur, is oxidised to  $\text{SO}_2$
- **$\text{SO}_2$** : **tracer for atmospheric dynamics and metallicity** of giant planets

Feinstein,Booth,Bergner+2025

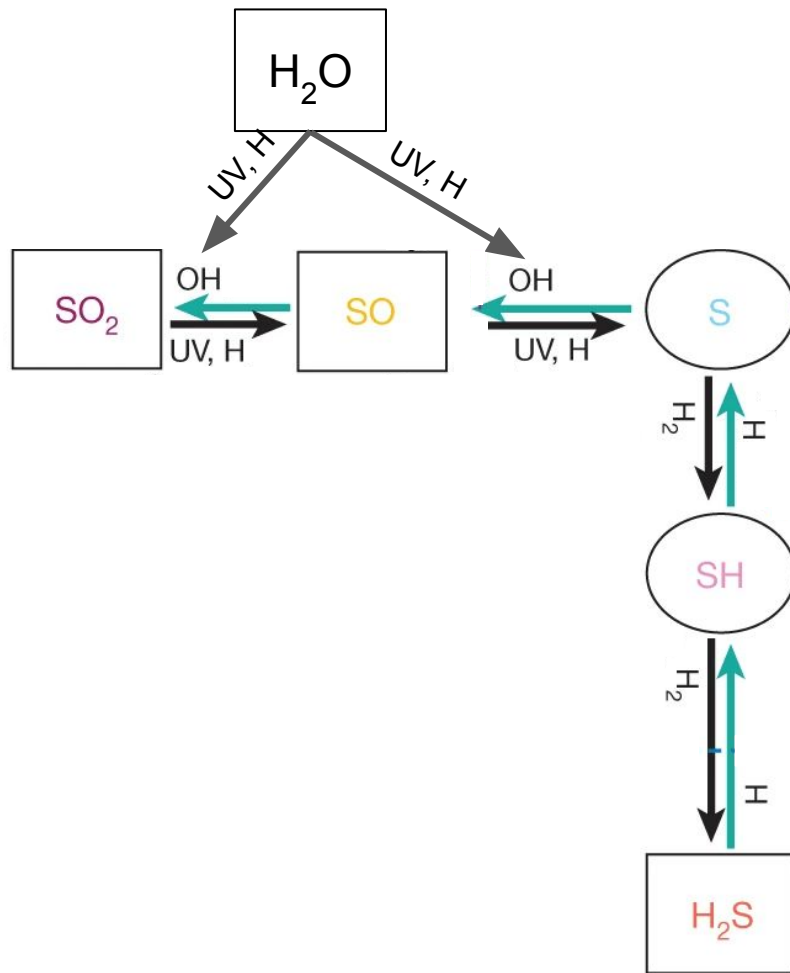


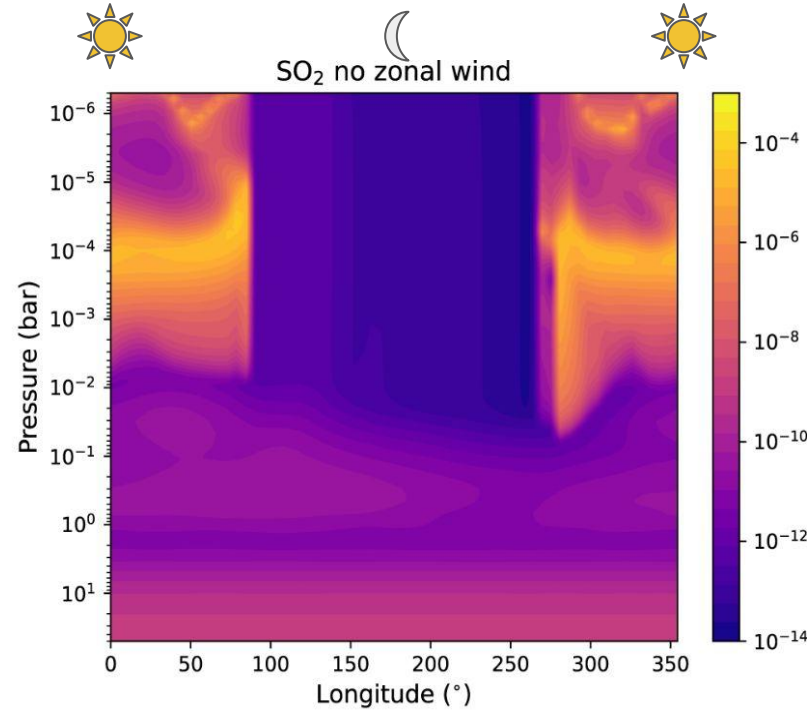
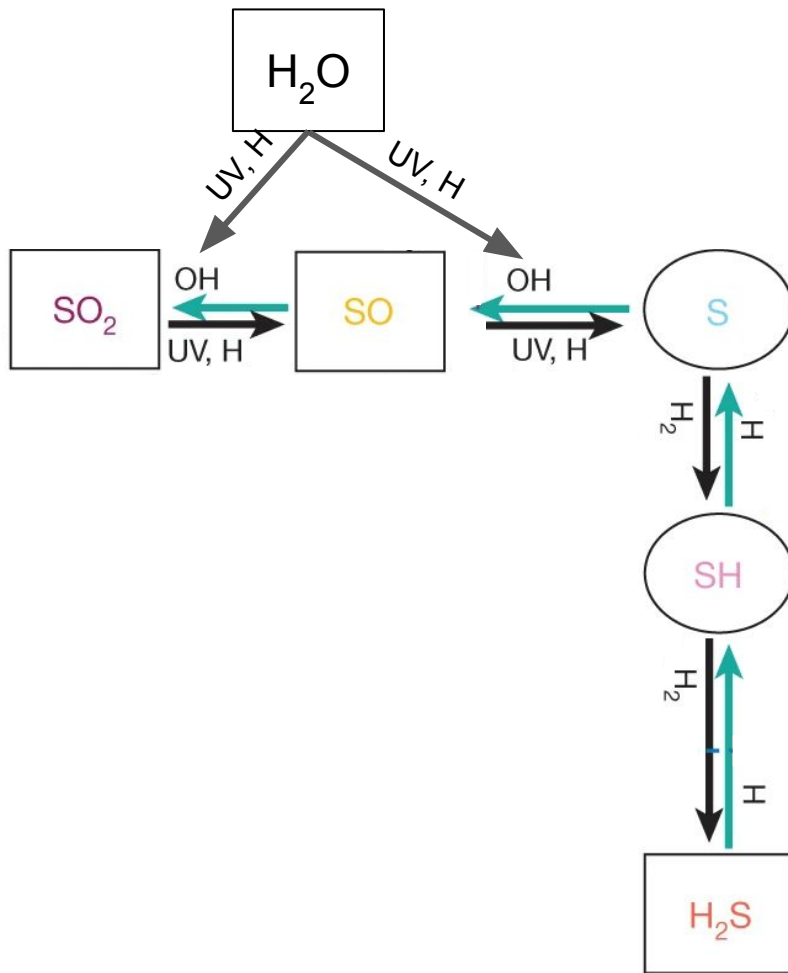




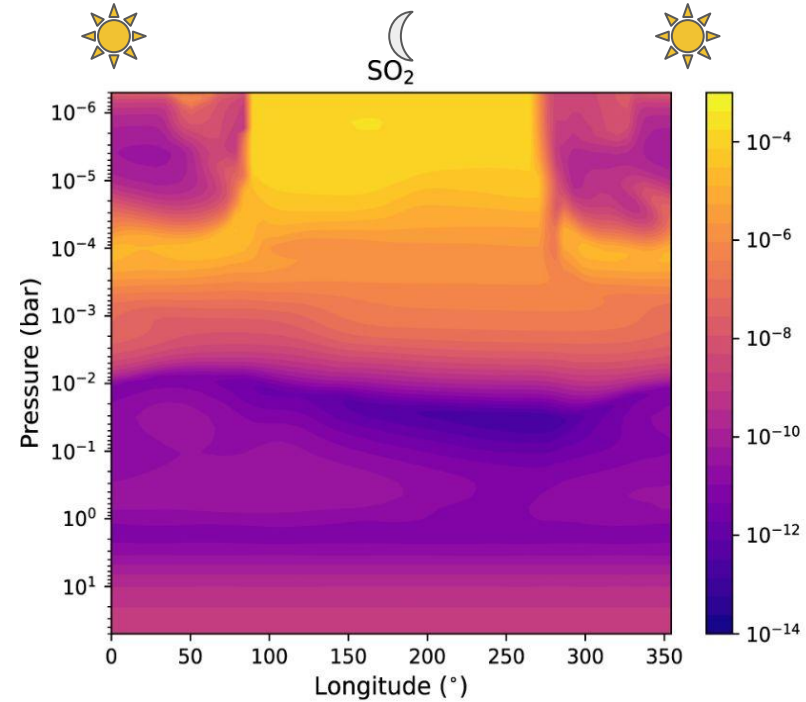
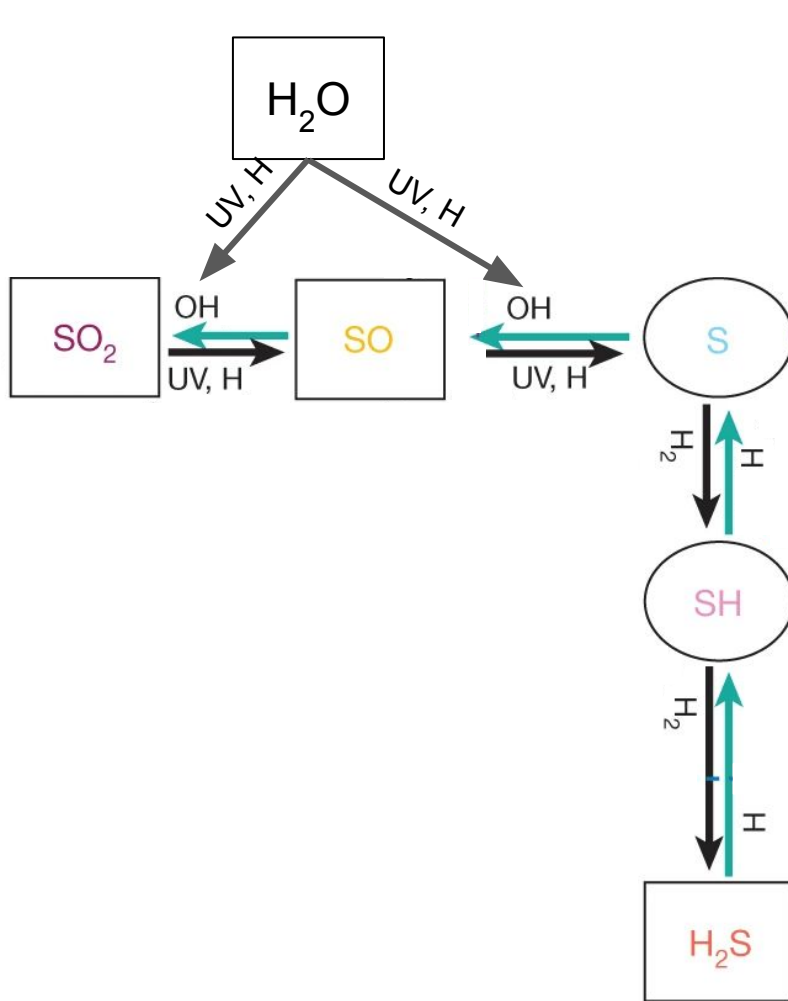




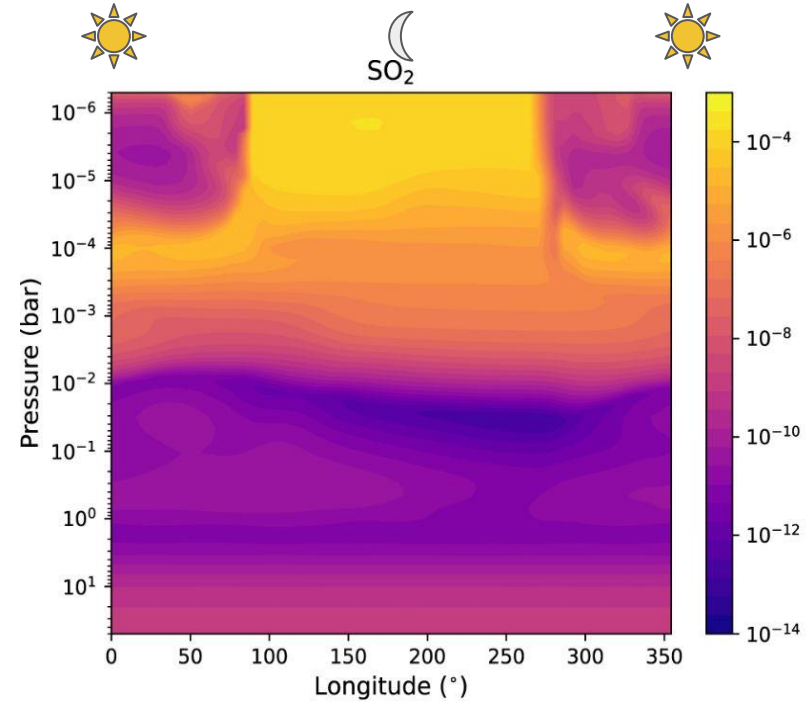
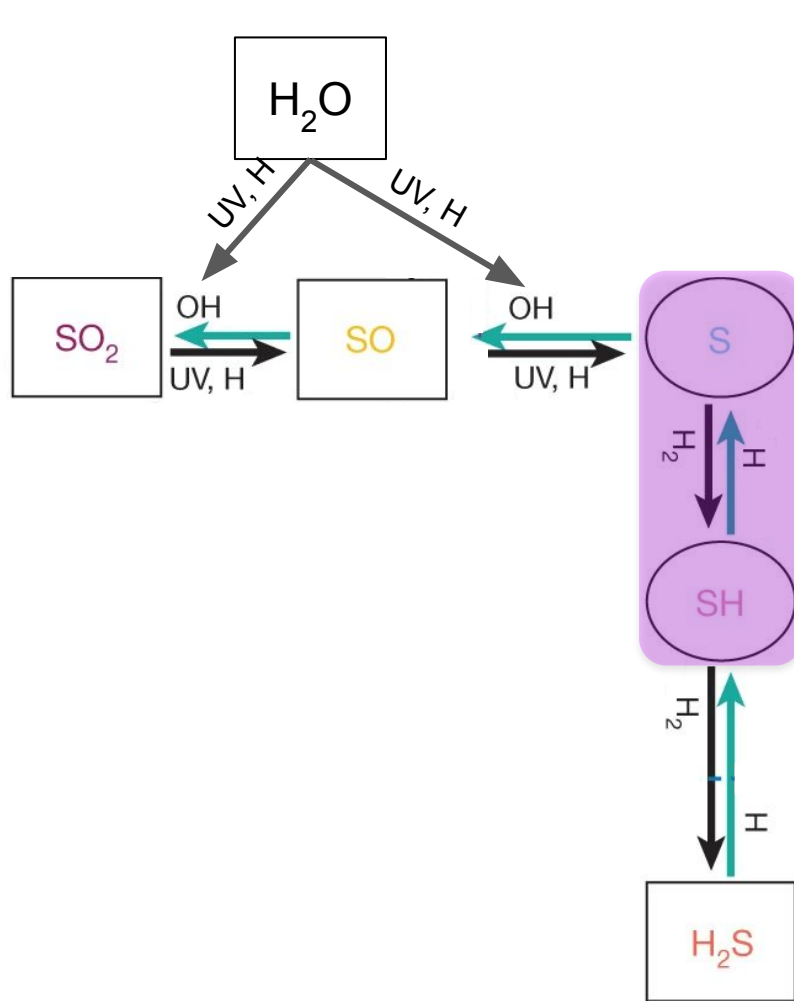




**No horizontal transport**



**With horizontal transport**





**Table 4.** Sulfur Reaction Rate Coefficient Data Needed for Exoplanet Studies

Reaction <sup>a</sup>		Notes <sup>b</sup>
S + S <sub>3</sub>	→ S <sub>2</sub> + S <sub>2</sub>	Need data
S + S <sub>4</sub>	→ S <sub>2</sub> + S <sub>3</sub>	Need data
S + SH	→ S <sub>2</sub> + H	Some data exist; need accurate values and T dependence
S + HS <sub>2</sub>	→ S <sub>2</sub> + SH	Sendt et al. (2002); need extension to lower & higher T
S + CO + M	→ OCS + M	† Need low and high pressure limits and their T dependence
S + CS + M	→ CS <sub>2</sub> + M	† Need low and high pressure limits and their T dependence
S + HCS	→ CS <sub>2</sub> + H	† Alzueta et al. (2019) estimate; need confirmation
S <sub>2</sub> + H + M	→ HS <sub>2</sub> + M	Sendt et al. (2002); need confirmation and extension to lower T
S <sub>2</sub> + S <sub>2</sub> + M	→ S <sub>4</sub> + M	† Some data exist; need accurate values and T dependence
S <sub>4</sub> + S <sub>4</sub> + M	→ S <sub>8</sub> + M	† Need low and high pressure limits and their T dependence
SH + H	→ H <sub>2</sub> + S	Some data exist; need accurate values and T dependence
SH + H + M	→ H <sub>2</sub> S + M	Need low and high pressure limits and T dependence
SH + S <sub>3</sub>	→ HS <sub>2</sub> + S <sub>2</sub>	Need data
SH + S <sub>4</sub>	→ HS <sub>2</sub> + S <sub>3</sub>	Need data
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SH + H <sub>2</sub> S <sub>2</sub>	→ H <sub>2</sub> S + HS <sub>2</sub>	Sendt et al. (2002); need extension to lower T
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HS <sub>2</sub> + H	→ SH + SH	Sendt et al. (2002); need confirmation
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HS <sub>2</sub> + HS <sub>2</sub>	→ H <sub>2</sub> S <sub>2</sub> + S <sub>2</sub>	Sendt et al. (2002); need extension to lower T
HS <sub>2</sub> + CS	→ CS <sub>2</sub> + SH	Need data and T dependence
CS + SO	→ CO + S <sub>2</sub>	Need data and T dependence
CS + SO	→ OCS + S	Need data and T dependence
HOSO + H	→ products	Need data and product branching ratios
HOSO + O	→ products	Need data and product branching ratios
HOSO + OH	→ products	Need data and product branching ratios
HOSO + S	→ products	Need data and product branching ratios
HOSO + SH	→ products	Need data and product branching ratios
S <sub>2</sub> O + H	→ SO + OH	Need data

<sup>a</sup> M represents any third body such as the dominant background gas.<sup>b</sup> † represents particularly pressing need.

Need  
lab  
data!

# Many reactions involving C-S bond

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HS <sub>2</sub> + HS <sub>2</sub>	→ H <sub>2</sub> S <sub>2</sub> + S <sub>2</sub>	Sendt et al. (2002); need extension to lower T
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<sup>a</sup> M represents any third body such as the dominant background gas.

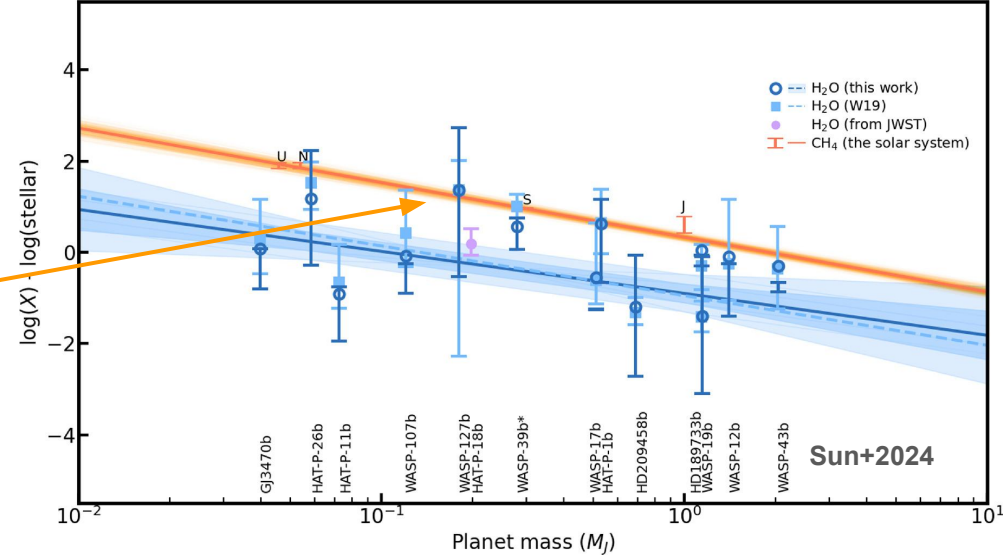
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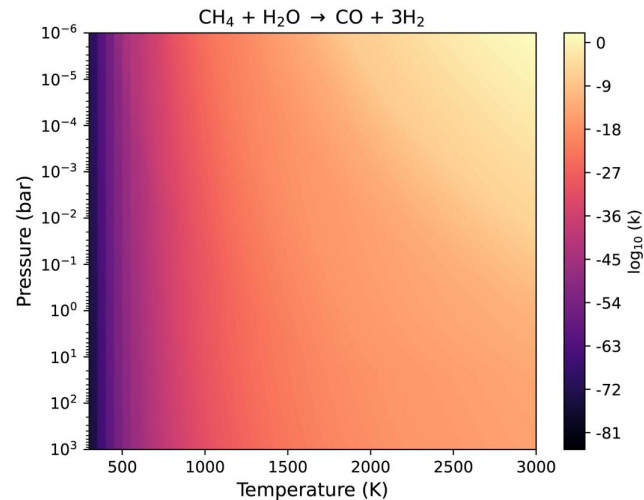
S chemistry is affected by C chemistry,  
and one of the key C species in giant  
planet atmospheres is CH<sub>4</sub>

# Why care about CH<sub>4</sub>?

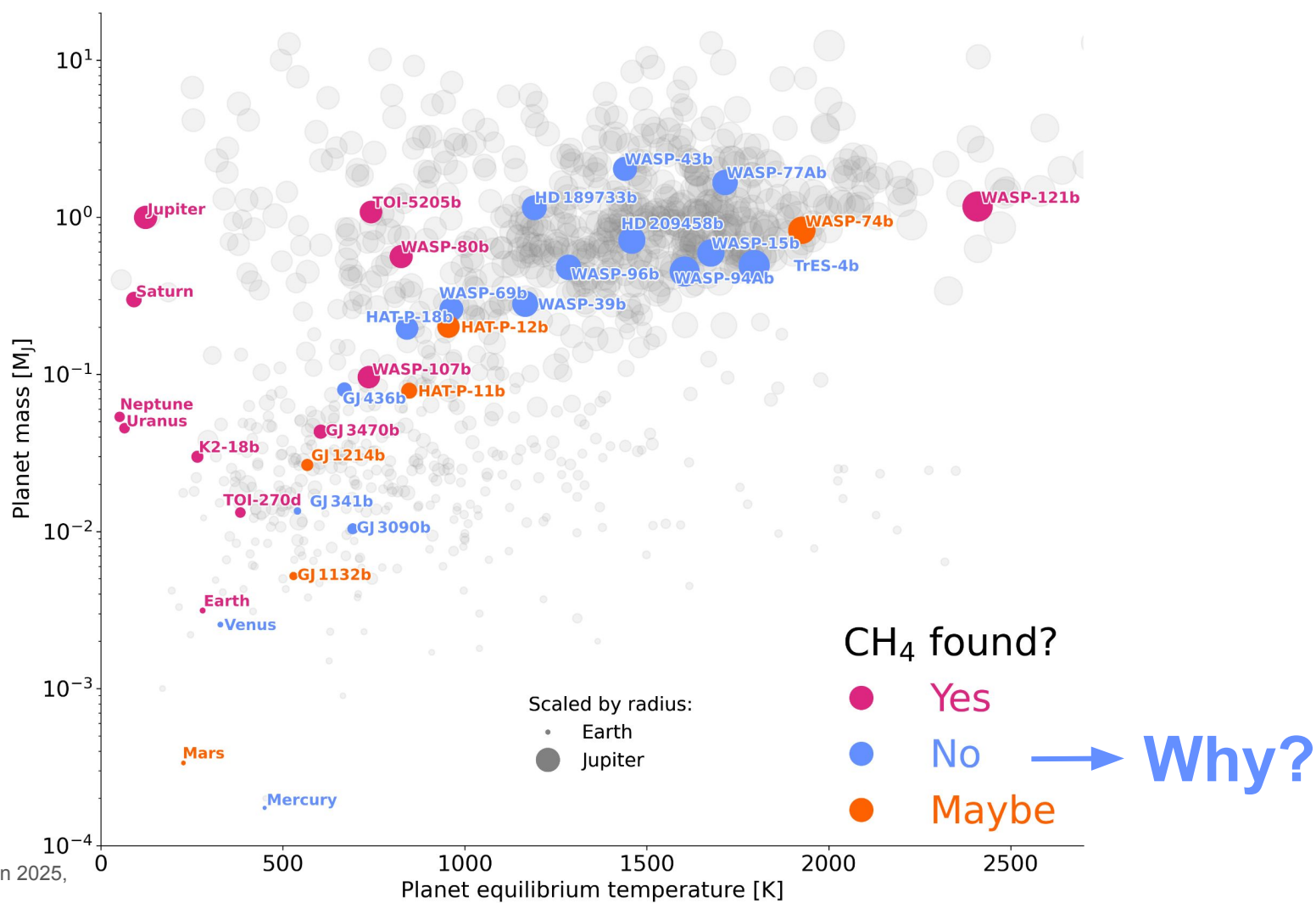
- CH<sub>4</sub> is a **proxy for metallicity** for the **solar system giants**
- CH<sub>4</sub> is a **haze precursor**
- CH<sub>4</sub> depletion as a **proxy for the deep atmosphere temperature**

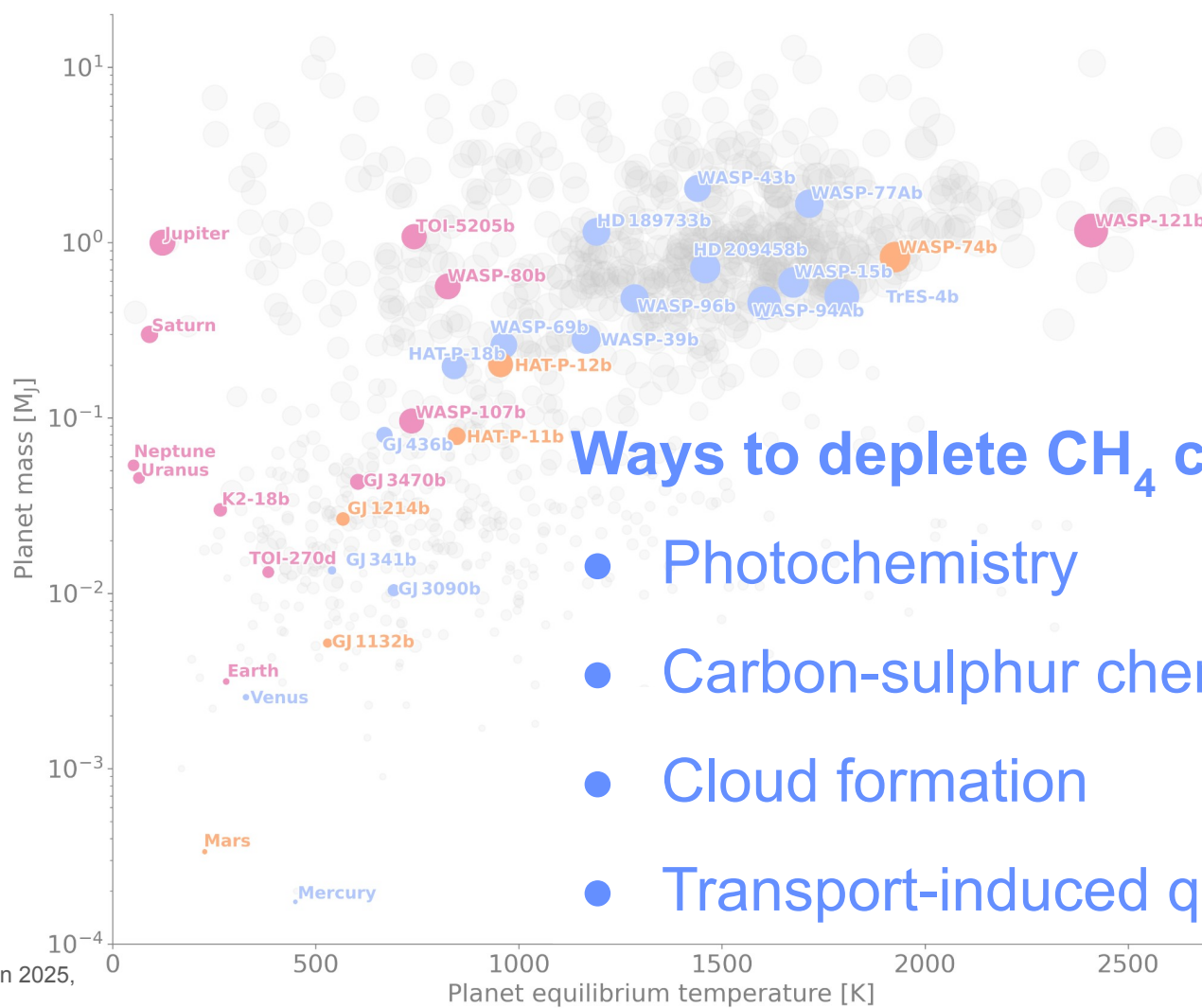


- CH<sub>4</sub> holds **most carbon at low temperatures**, while CO dominates at high temperatures
- But: CH<sub>4</sub>-CO transition is smooth







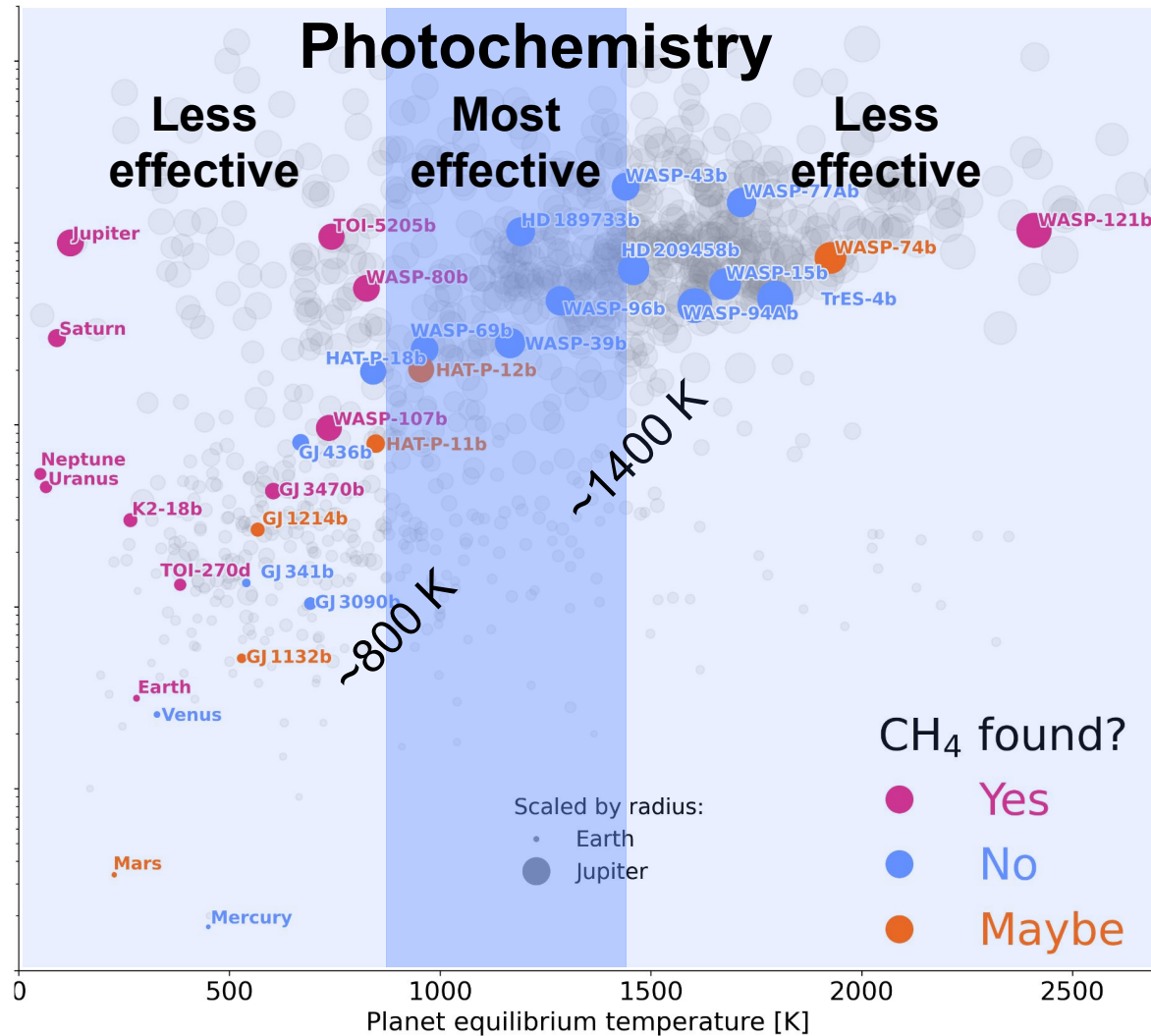


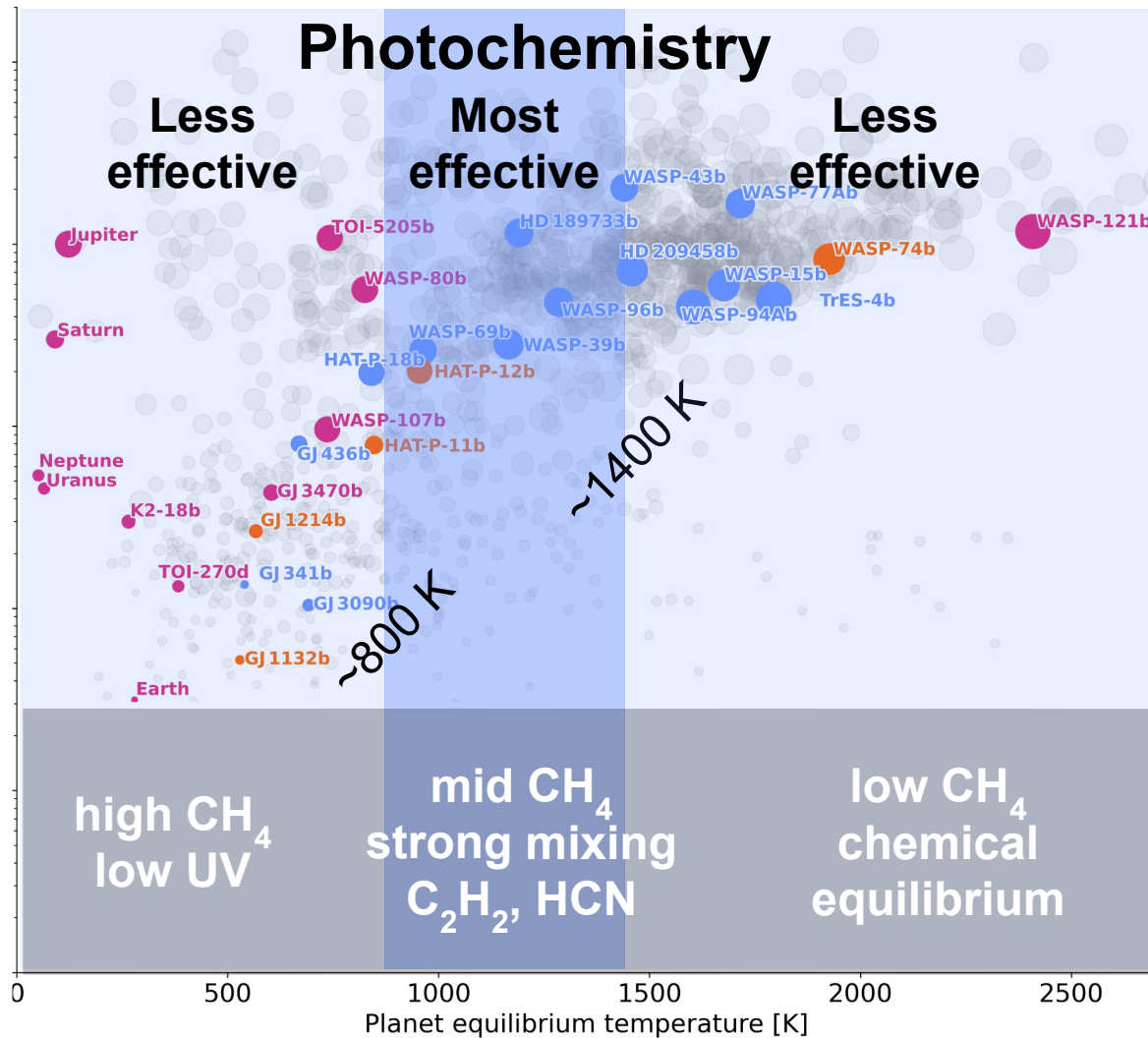
# Photochemistry

Less  
effective

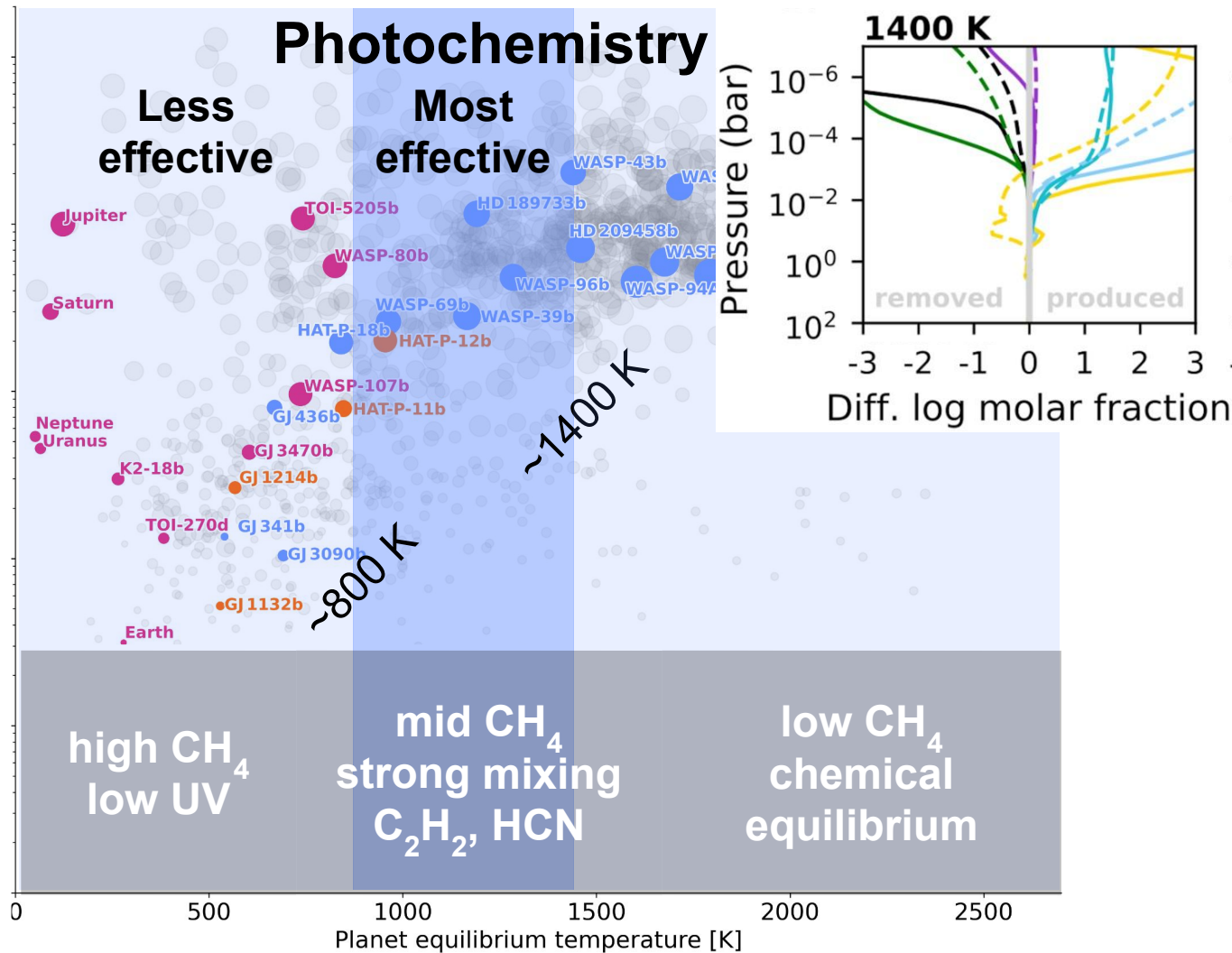
Most  
effective

Less  
effective





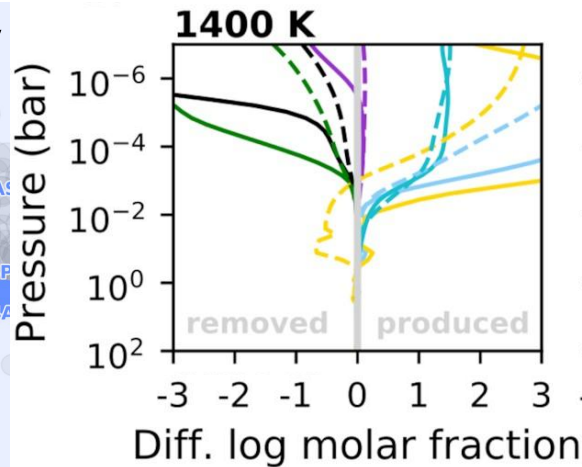
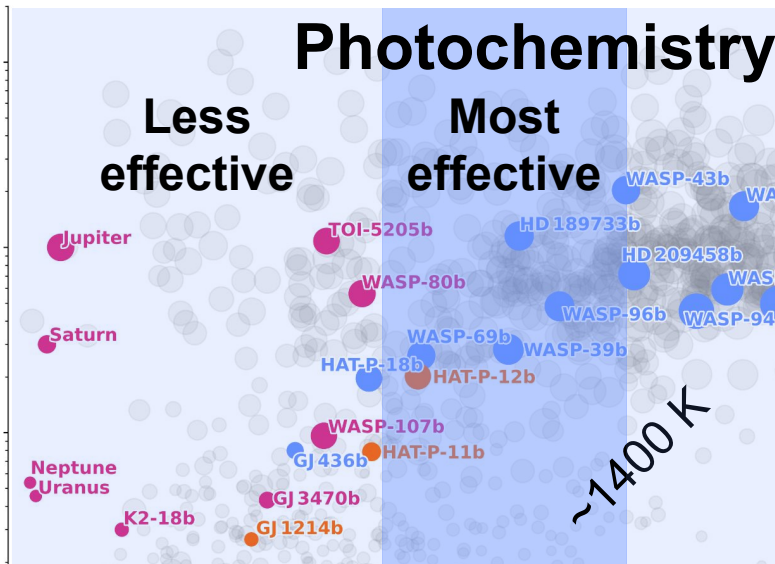




# Photochemistry

Less effective

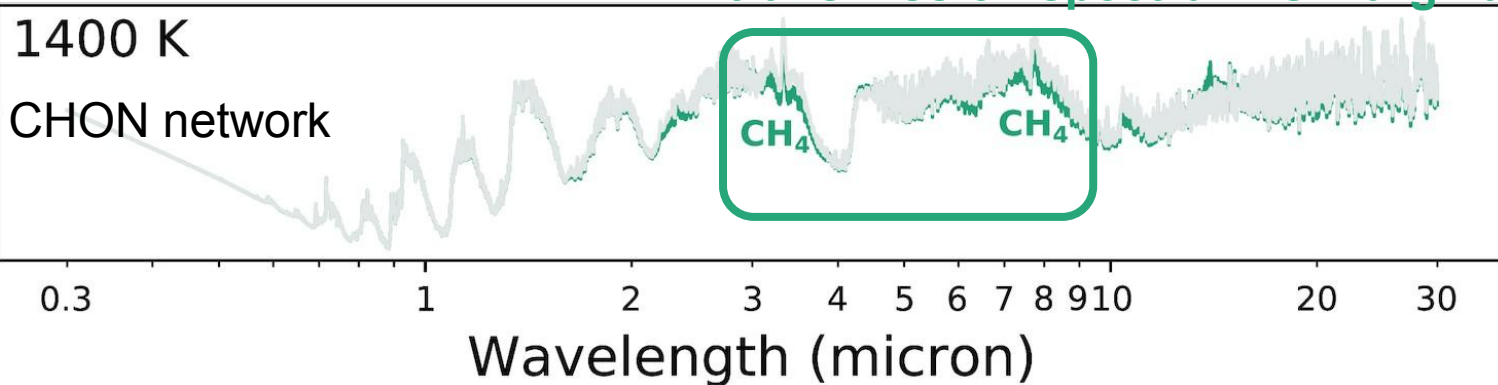
Most effective



**CH<sub>4</sub> is depleted but the effect on the transmission spectrum is marginal**

Transit radius ( $R_{Jup}$ )

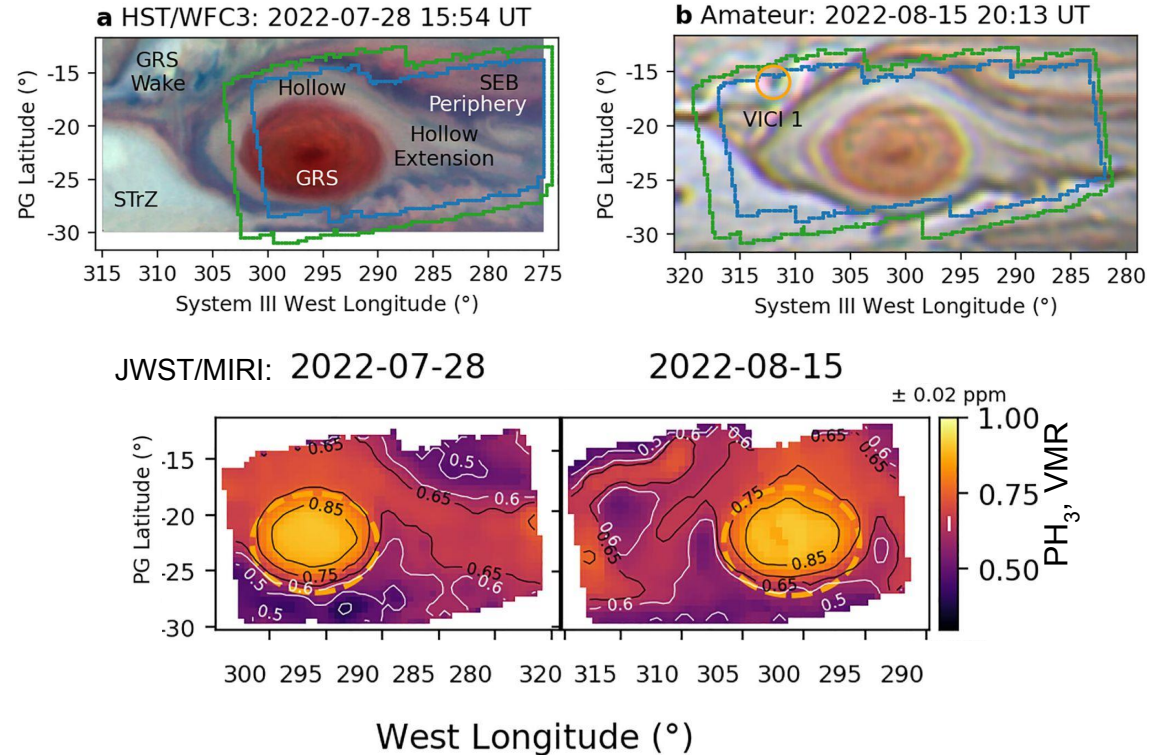
— with photochemistry  
— without photochemistry



# Lessons from the solar system:

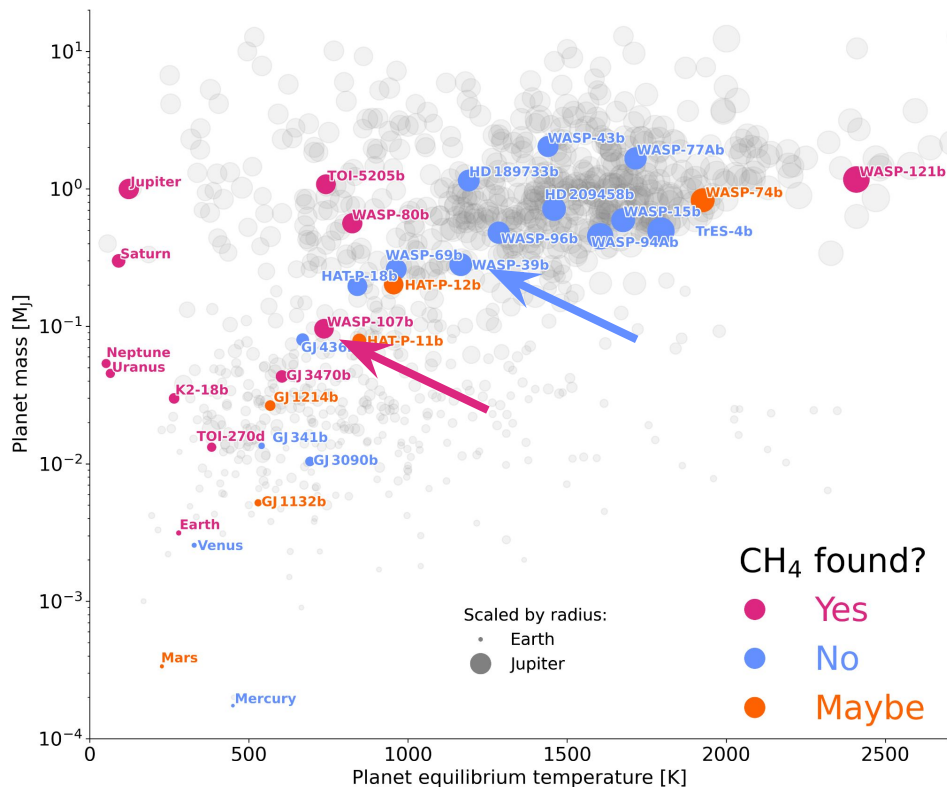
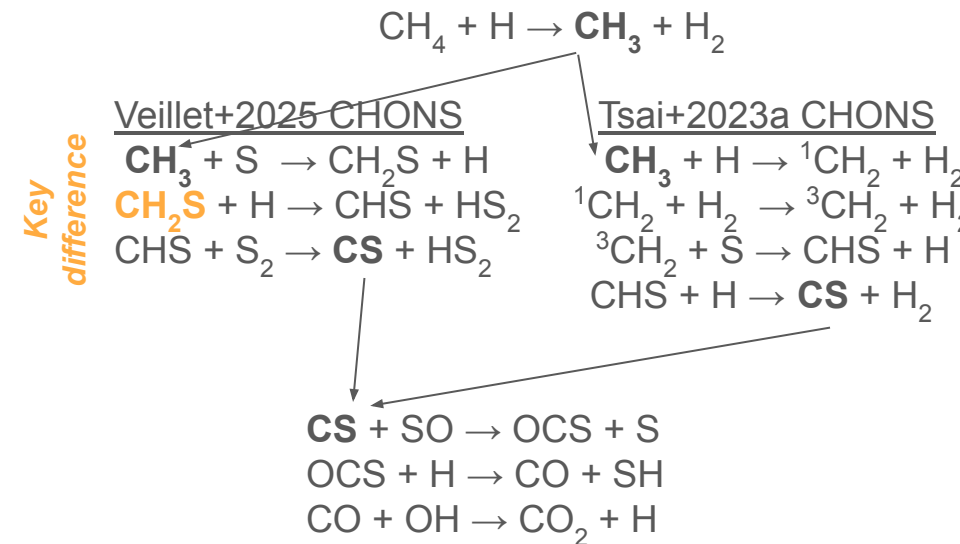
## Phosphine in Jupiter's Great Red Spot

- Excess in  $\text{PH}_3$  and aerosols coincides with still unidentified red chromophore
- **Aerosols shield  $\text{PH}_3$  from UV** in this long-lived anticyclone



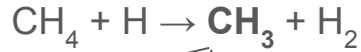
# CH<sub>4</sub> depletion due to: Carbon-sulphur chemistry

Carbon-sulphur chemistry depletes CH<sub>4</sub> & does it differently in different chemical networks. E.g.:

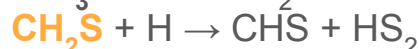


# CH<sub>4</sub> depletion due to: Carbon-sulphur chemistry

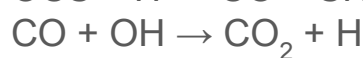
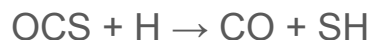
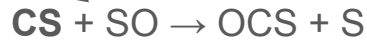
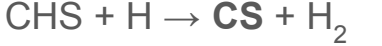
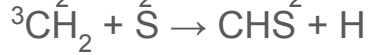
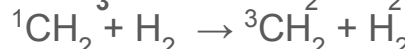
Carbon-sulphur chemistry depletes CH<sub>4</sub> & does it differently in different chemical networks. E.g.:



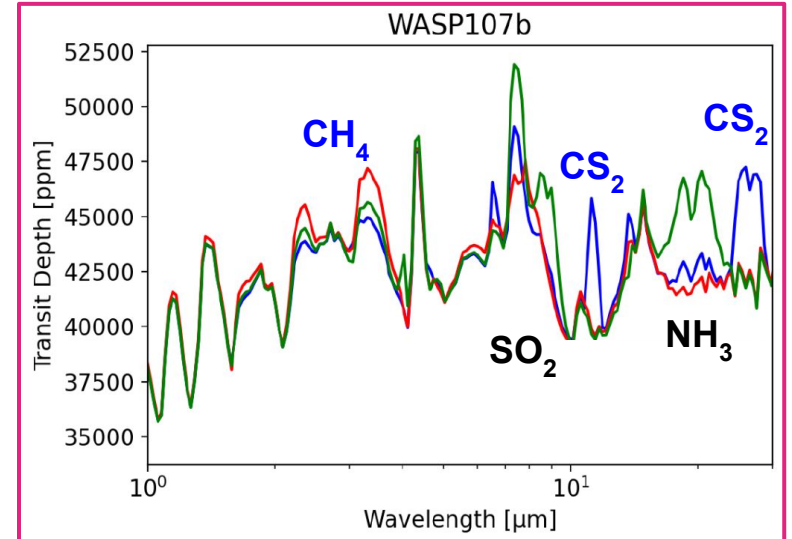
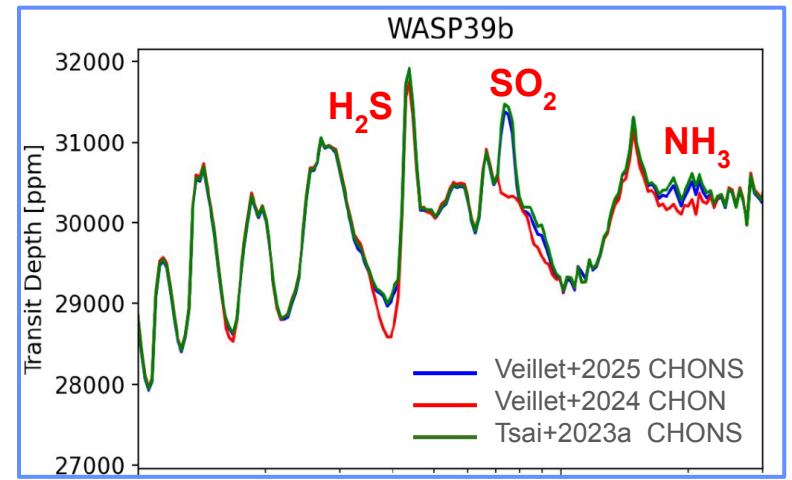
Veillet+2025 CHONS



Tsai+2023a CHONS



Key  
difference



# Aside:

## Families of chemical networks for giant planets

Last updated	C	H	O	N	S	P	Cl	Model	Species	Notes
Moses+2013	✓	✓	✓	✓	✗	✗	✗	KINETICS	92	originally developed for solar system
Hu+2015	✓	✓	✓	✓	✓	✗	✗	MEAC	111	originally developed for rocky exoplanets
Gao+2016, Johnson+2022	✓	✓	✓	✓	✓	✗	✗	EPACRIS	varies	computer-aided chemical reaction network generator
Hobbs2021	✓	✓	✓	✓	✓	✗	✗	LEVI	a lot	STAND
Tsai+2021	✓	✓	✓	✓	✓	✗	✗	VULCAN	96	open source, widely used
Tsai+2022	✓	✓	✓	✓	✗	✗	✗	Exo-FMS	12	MINI-CHEM
Lee+2024	✗	✓	✓	✗	✗	✓	✗	VULCAN	32	phosphorus
Wogan+2024	✓	✓	✓	✓	✓	✗	✓	Photochem	111	Zahnle, chlorine
Veillet+2025	✓	✓	✓	✓	✓	✗	✗	FRECKLL	226	verified against combustion experiments

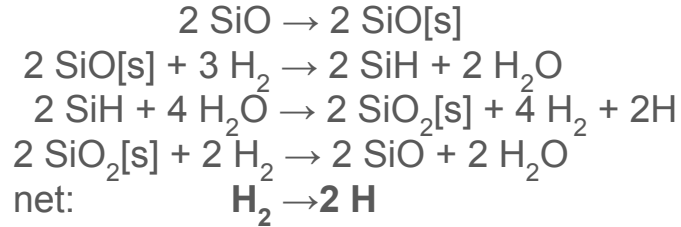
Sorry if I missed your “family”! Please let me know.

**New laboratory & ab initio data are needed!**

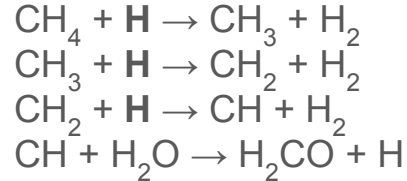


# CH<sub>4</sub> depletion due to: Cloud formation

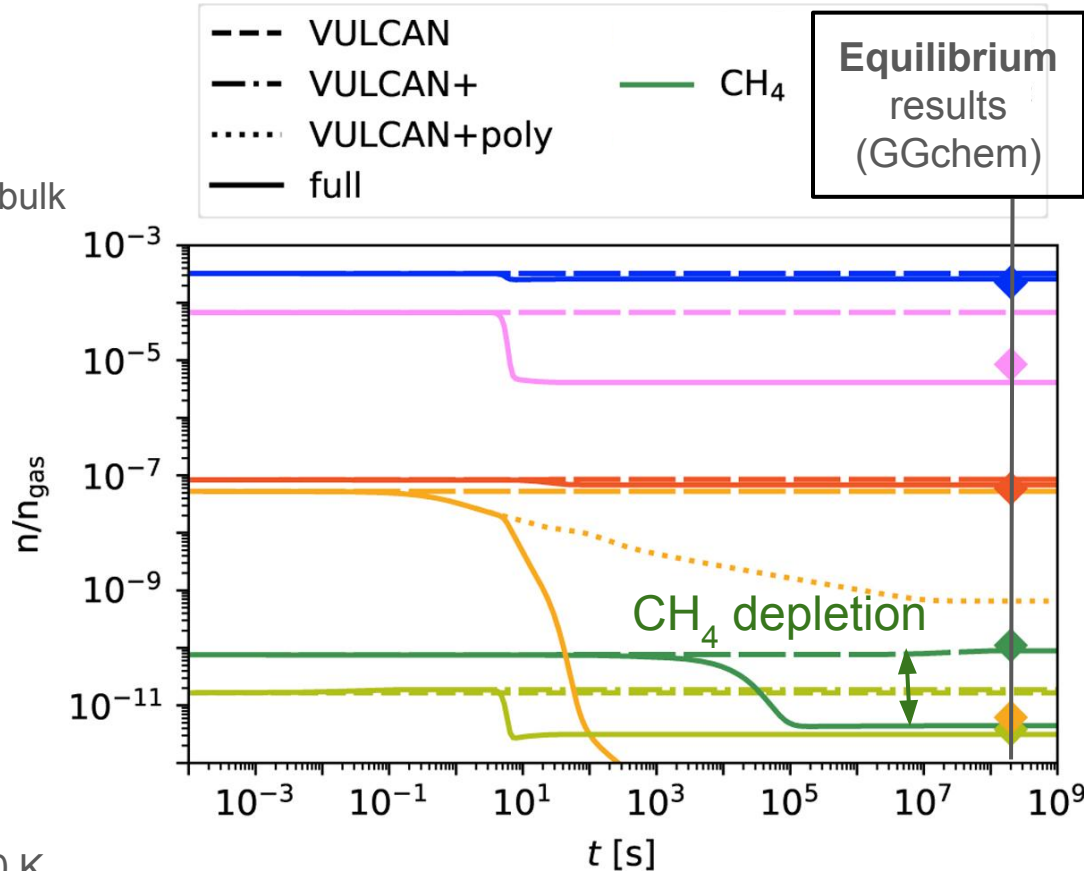
- Kinetic cloud formation model
- Cycling between surface reactions of cloud bulk growth:



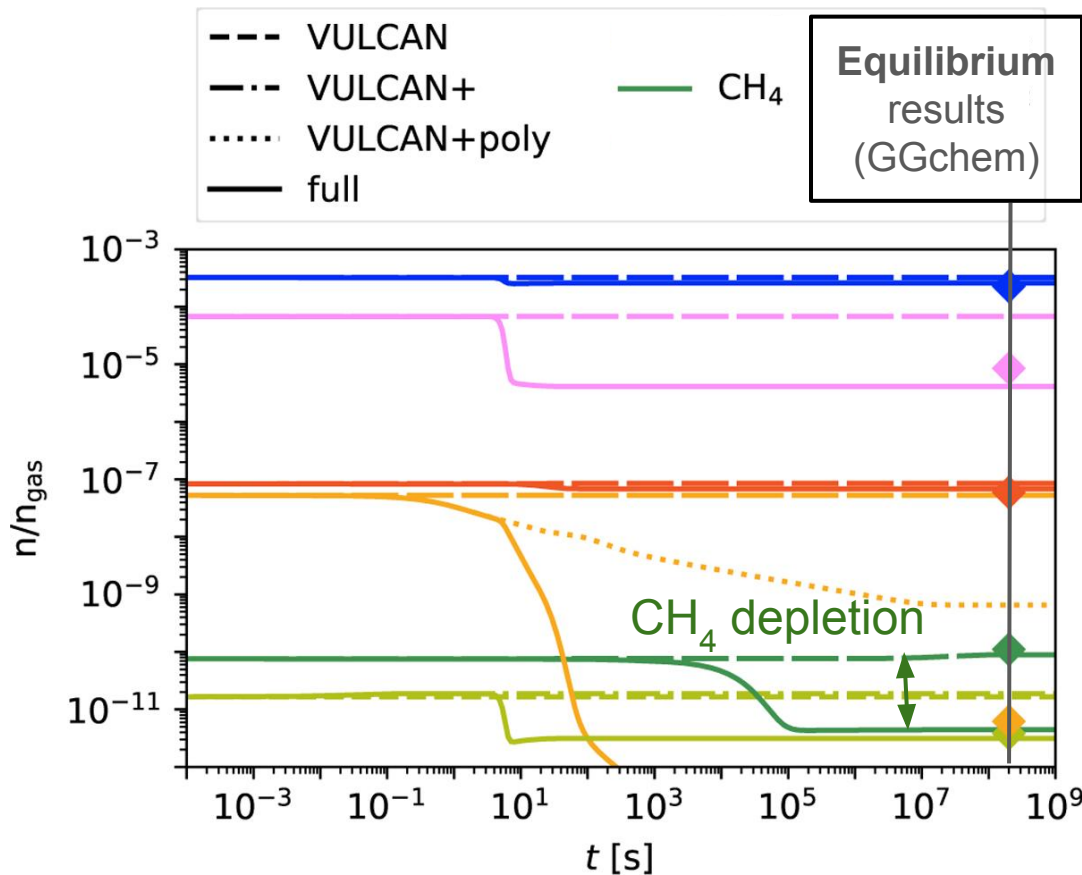
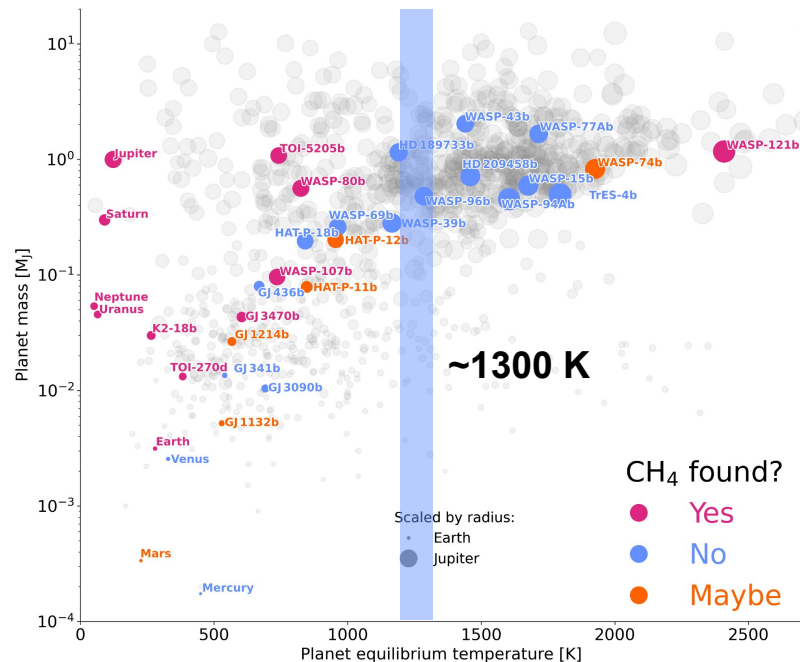
- Depletion via:



- CH<sub>4</sub> changes largest at <10<sup>-3</sup> bar and ~1300 K

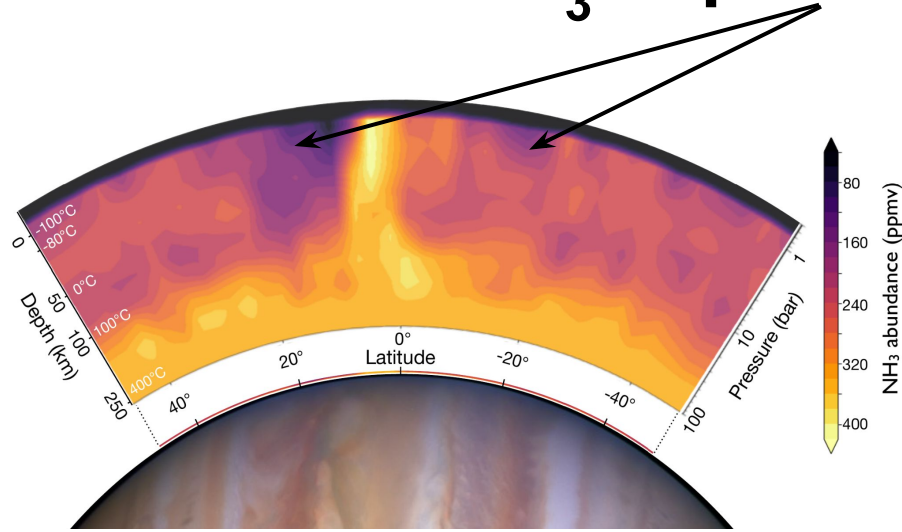


# CH<sub>4</sub> depletion due to: Cloud formation



# Lessons from the solar system:

## Storm-driven $\text{NH}_3$ depletion on Jupiter



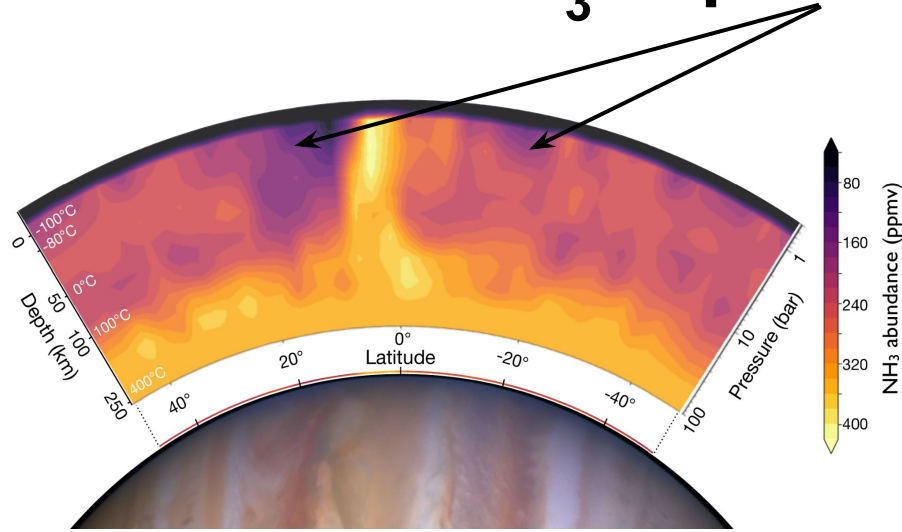
From a press release:

"Imke and I both were like, '**There's no way in the world this is true,**'" said Moeckel, ... "**So many things have to come together to actually explain this, it seems so exotic. I basically spent three years trying to prove this wrong. And I couldn't prove it wrong.**"

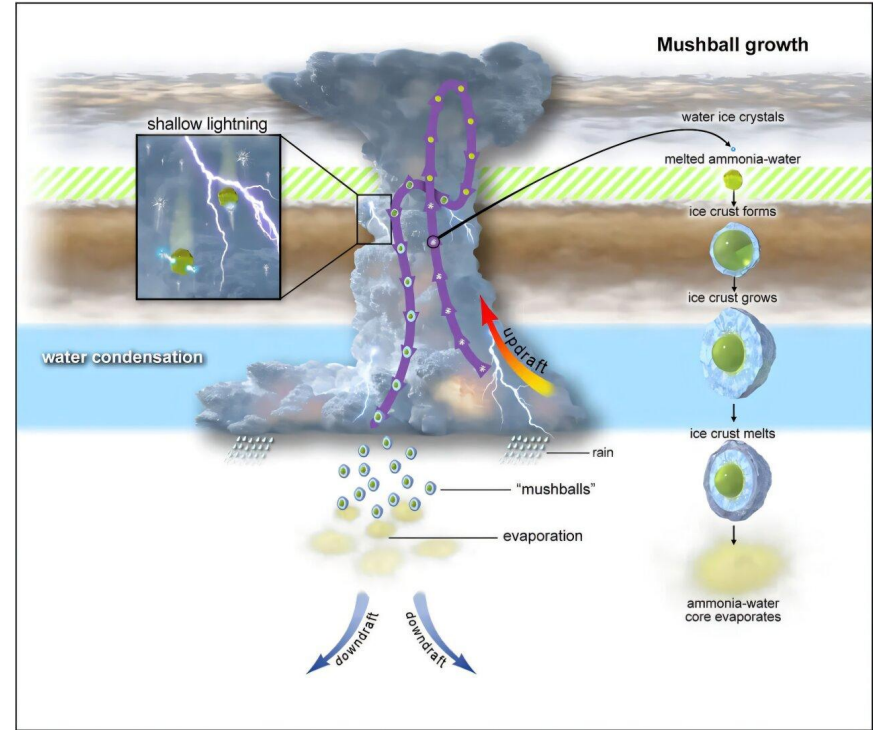
- Observation:  $\text{NH}_3$  depletion correlates with enhanced lightning activity

# Lessons from the solar system:

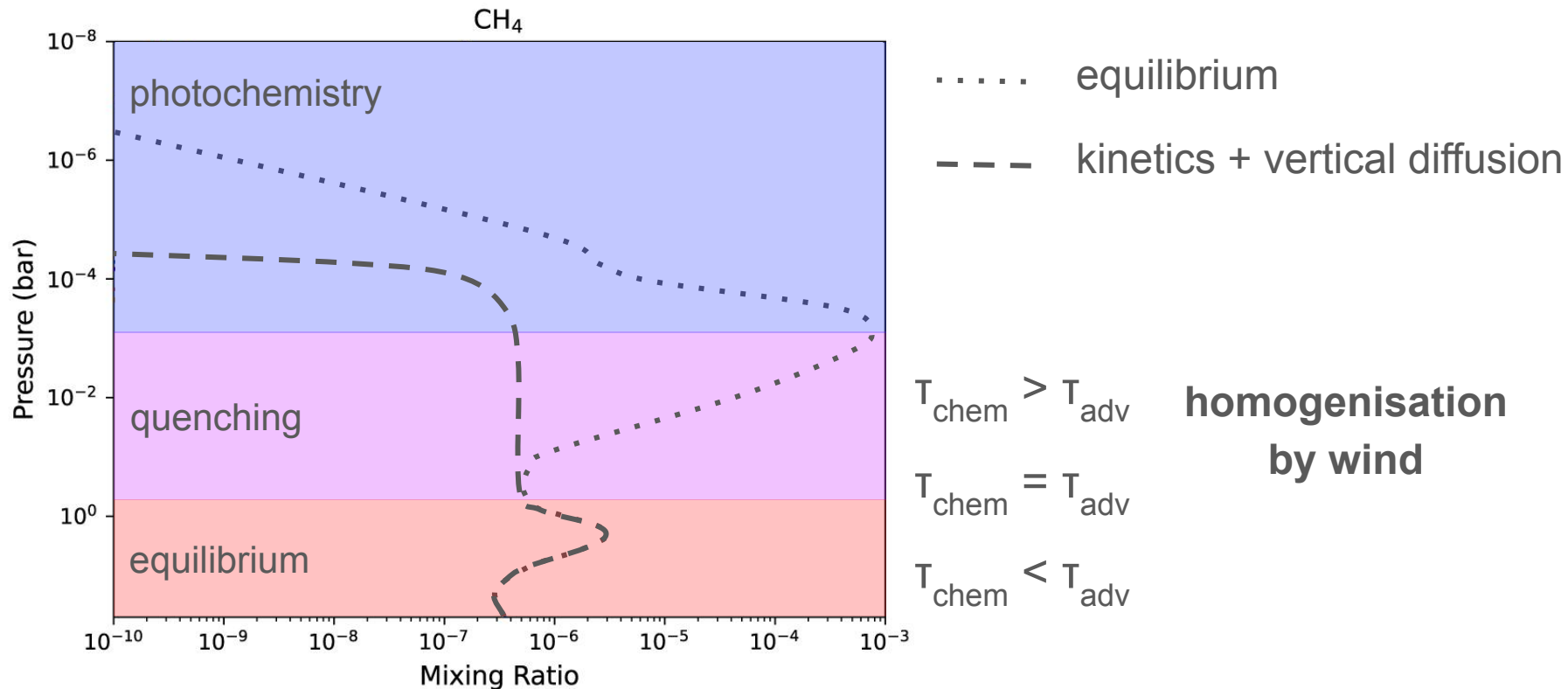
## Storm-driven $\text{NH}_3$ depletion on Jupiter



- Observation:  $\text{NH}_3$  depletion correlates with enhanced lightning activity
- Theory:  $\text{NH}_3$  vapour helps melt  $\text{H}_2\text{O}$  ice crystals at low temperatures ( $-85^\circ\text{C}$ ), forming  $\text{H}_2\text{O}$ - $\text{NH}_3$  mushballs
  - net effect: downward transport of  $\text{NH}_3$

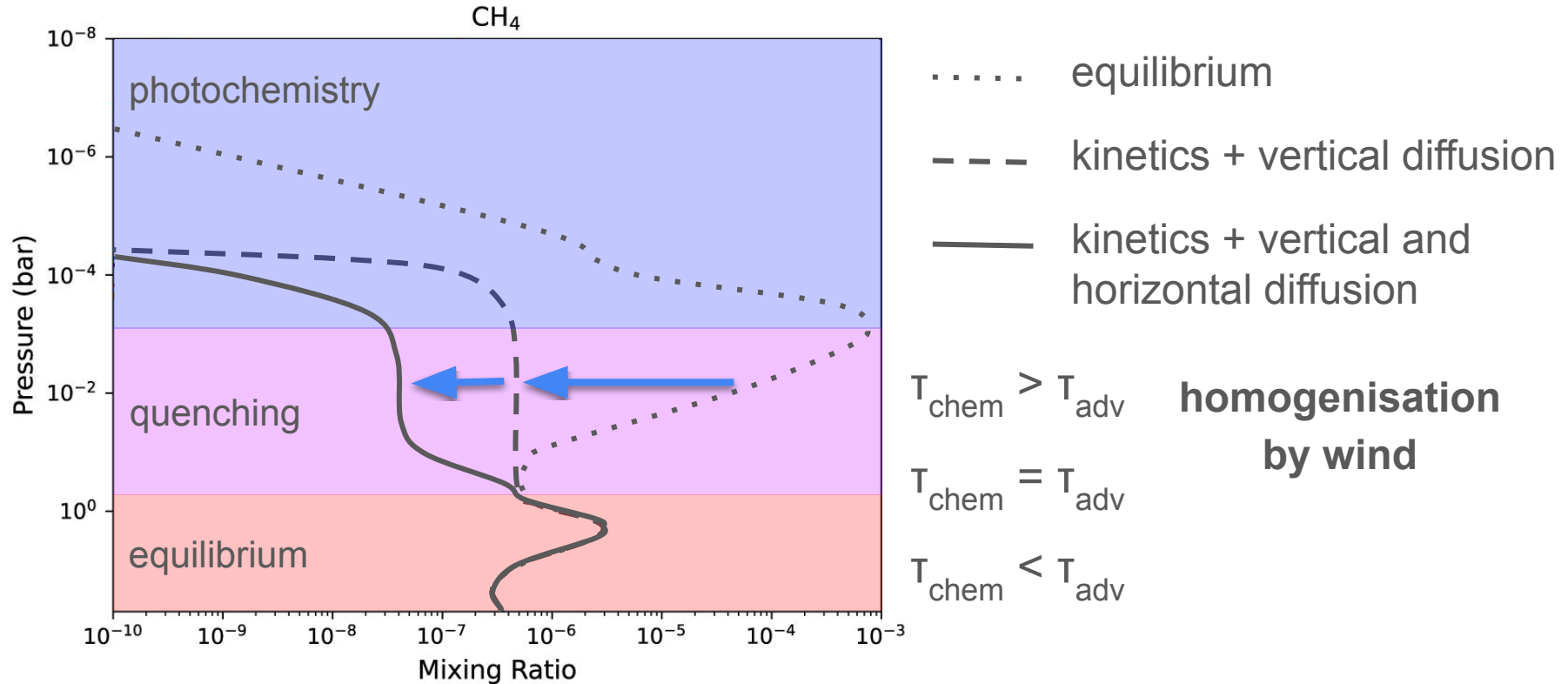


# CH<sub>4</sub> depletion due to: Transport-induced quenching

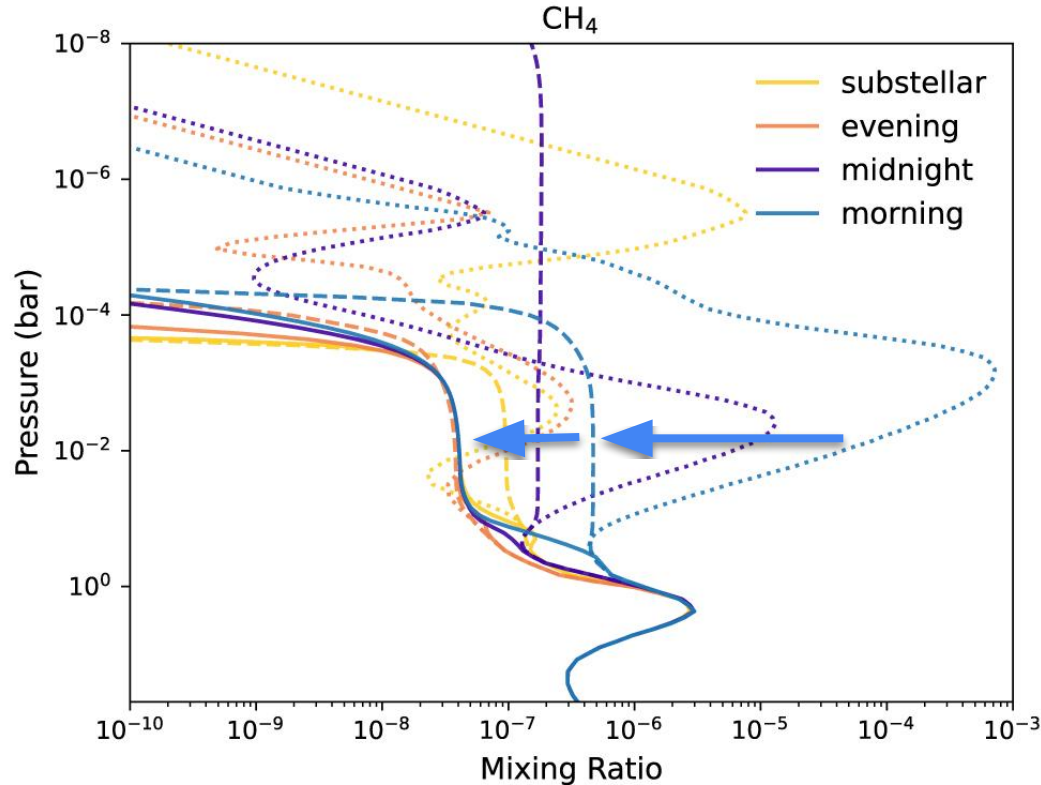




# CH<sub>4</sub> depletion due to: Transport-induced quenching



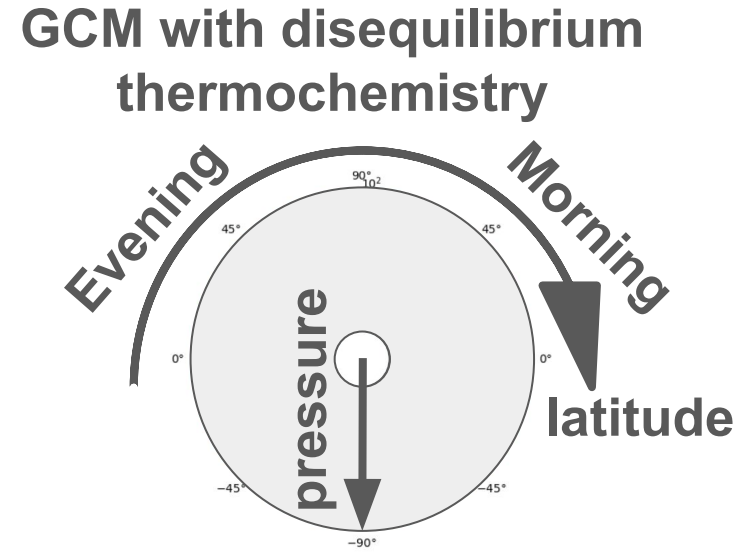
# CH<sub>4</sub> depletion due to: Transport-induced quenching



..... equilibrium  
--- kinetics + vertical diffusion  
— kinetics + vertical and horizontal diffusion

**Depletion by up to  
3-4 orders of  
magnitude!**

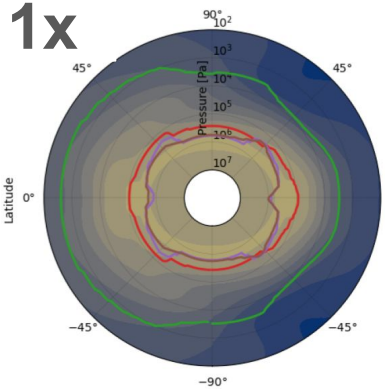
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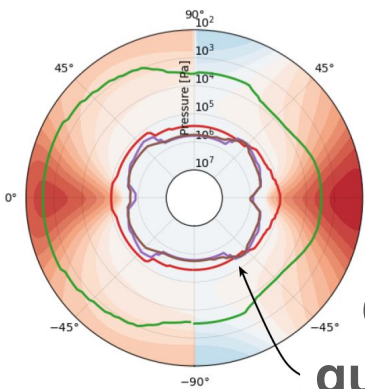
CH<sub>4</sub> depletion due to: **Transport-induced quenching**

**Temperature**

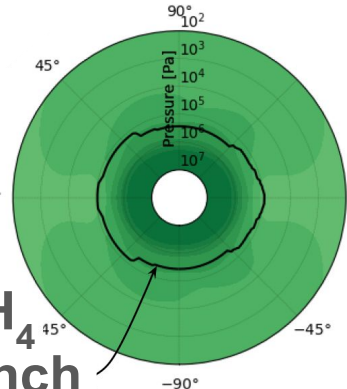
**1x**



**Zonal wind**

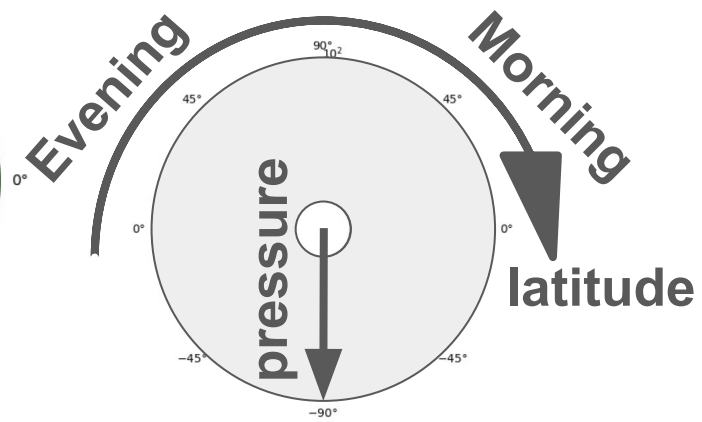


**CH<sub>4</sub>**



**CH<sub>4</sub> quench level**

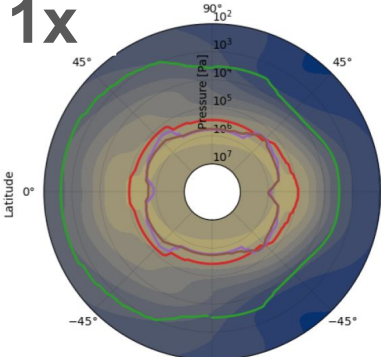
**GCM with disequilibrium thermochemistry**



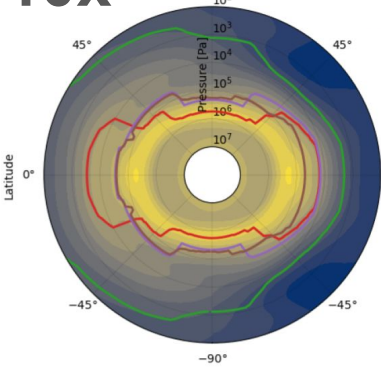
CH<sub>4</sub> depletion due to: **Transport-induced quenching**

**Temperature**

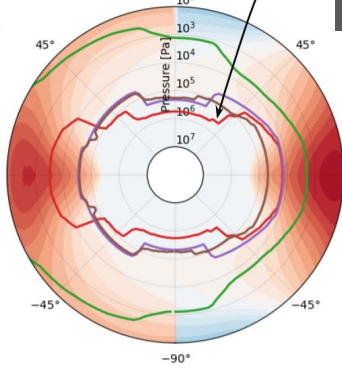
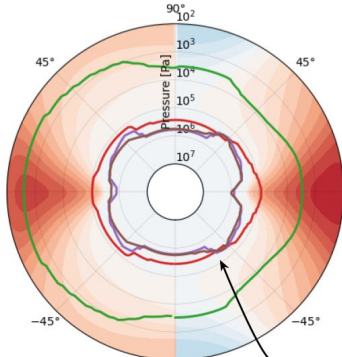
**1x**



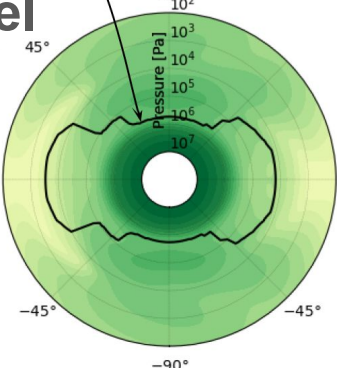
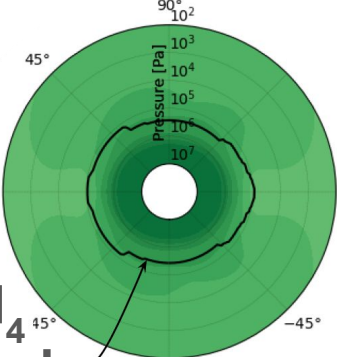
**10x**



**Zonal wind**

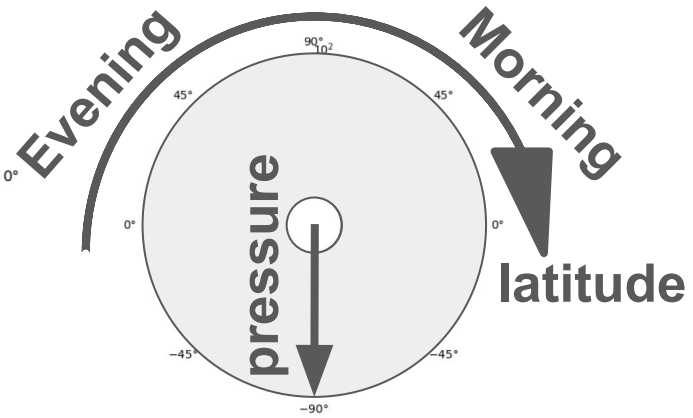


**CH<sub>4</sub>**



**CH<sub>4</sub> quench level**

**GCM with disequilibrium thermochemistry**





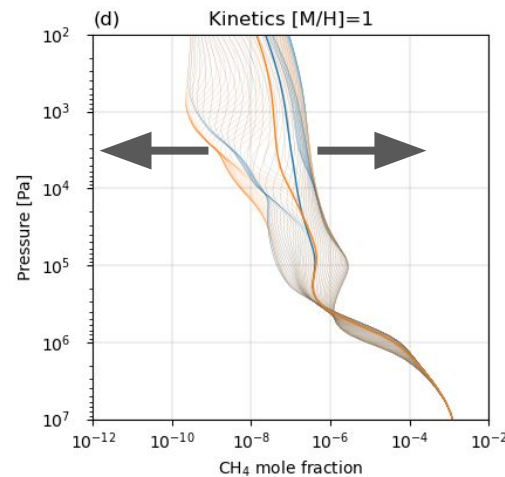
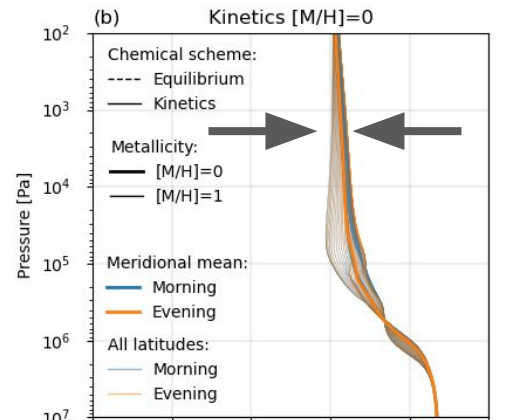
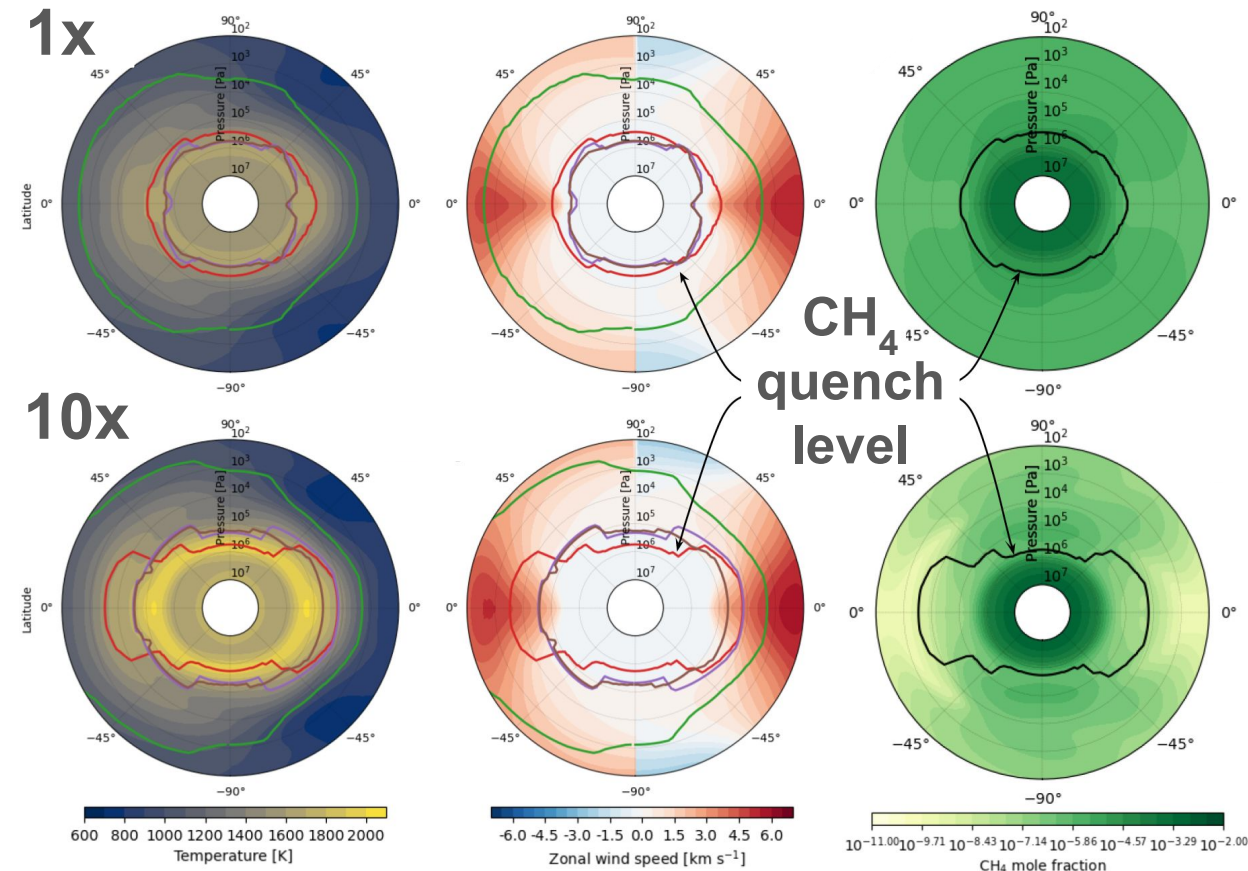
# CH<sub>4</sub> depletion due to: Transport-induced quenching

Temperature

Zonal wind

CH<sub>4</sub>

GCM with disequilibrium thermochemistry

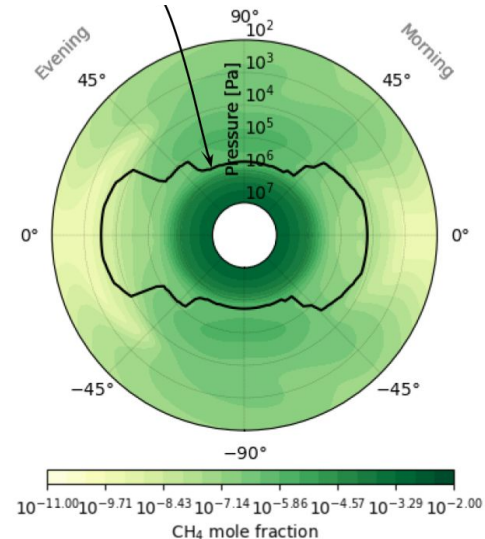
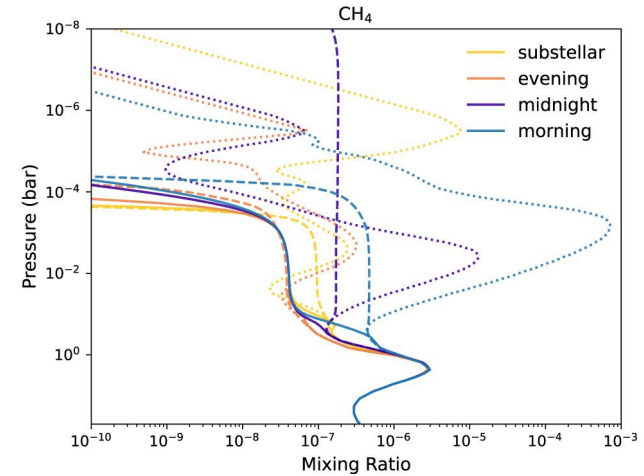


# Summary of: Transport-induced quenching

- Quenching - homogenisation of gas-phase composition by wind
- $\text{CH}_4$  is readily homogenised
- $\text{CH}_4$  is enhanced or depleted depending on where it quenches
  - $\text{CH}_4$  could be more depleted if it quenches inside the region of fast equatorial winds

## Implication:

- Be careful when using  $\text{CH}_4$  depletion as a deep atmosphere thermometer



# Summary

- Ever increasing wealth of chemical species detections allows for a detailed study of atmospheric processes
  - Keep moving from “just detections” to processes
- Keep increasing the complexity of disequilibrium chemistry models
  - “Atmospheres are not simple, one-dimensional constructs”, they are a highly coupled but beautiful mess
- Sulphur story is unraveling and there is more to come
- **Keep reporting non-detections**
  - CH<sub>4</sub> non-detections could be caused by photochemistry, C-S coupling, cloud formation, transport-induced quenching - **likely all together and more**
- Keep looking at the Solar System when interpreting extrasolar observations
- Need more chemical kinetics data