

Local and global impacts of C₁-C₃ alkyl nitrate photochemistry and emissions on tropospheric ozone

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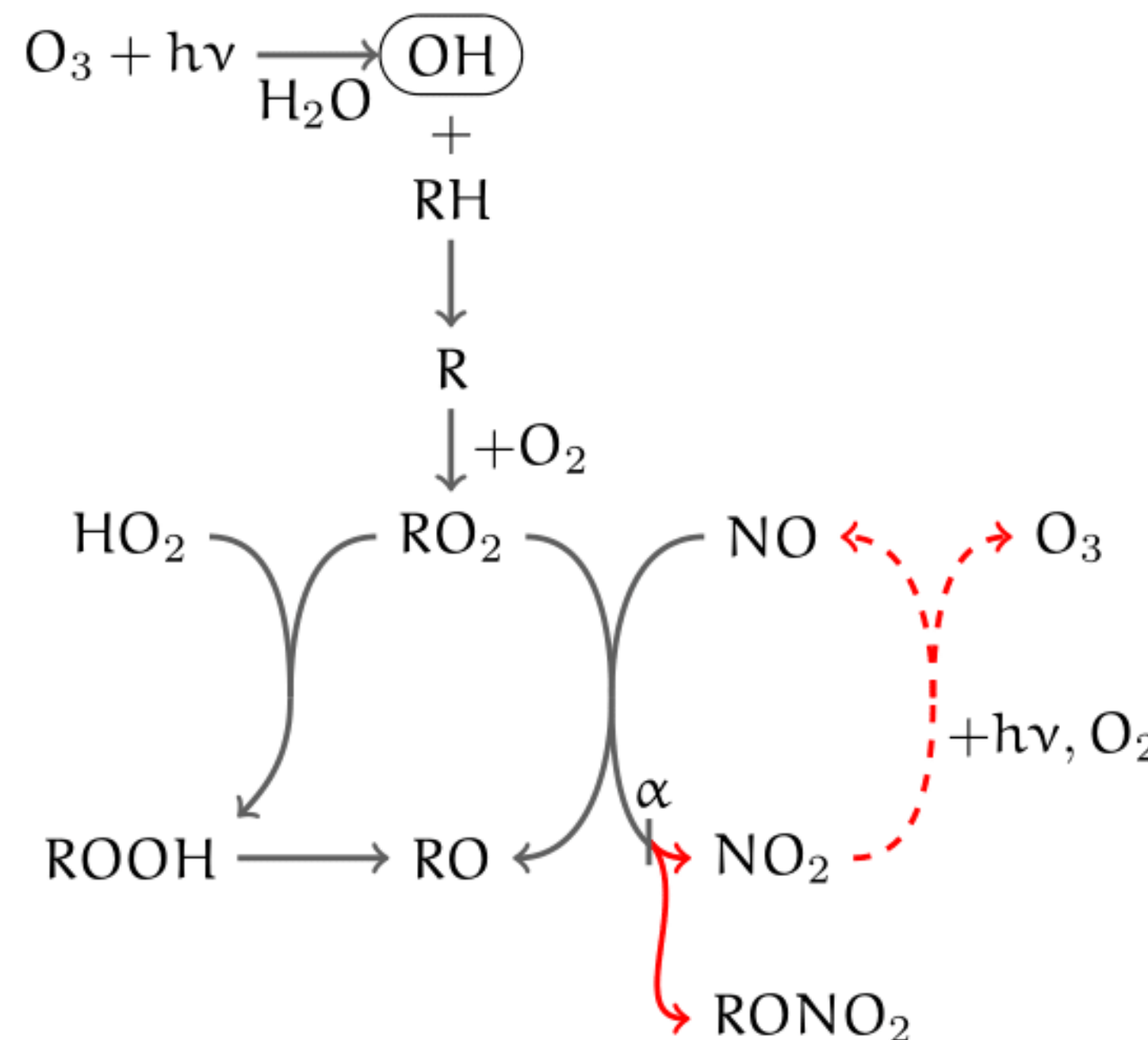
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Introduction

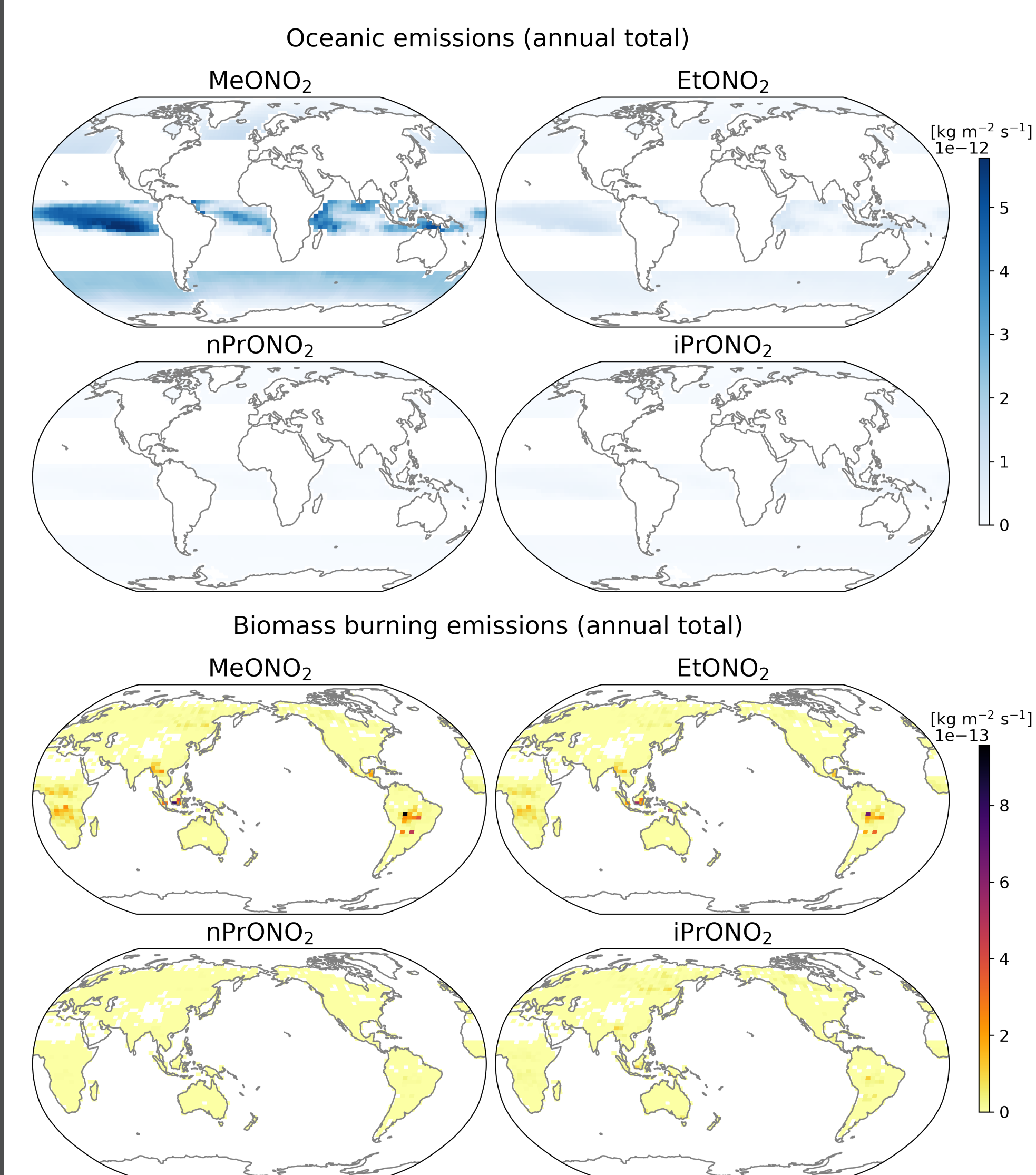
Alkyl nitrates (RONO₂) are directly emitted and photochemically produced from the oxidation of hydrocarbons in the presence of NO. Their formation terminates tropospheric O₃ production by temporarily storing the active form of nitrogen. Due to a relatively long lifetime of a few days to a few months, **RONO₂ can be destroyed far away from their sources by photolysis or OH oxidation, releasing NO₂ to the local atmosphere and altering O₃ concentrations.**



Few studies explored RONO₂ with a global chemistry-climate model. Here we:

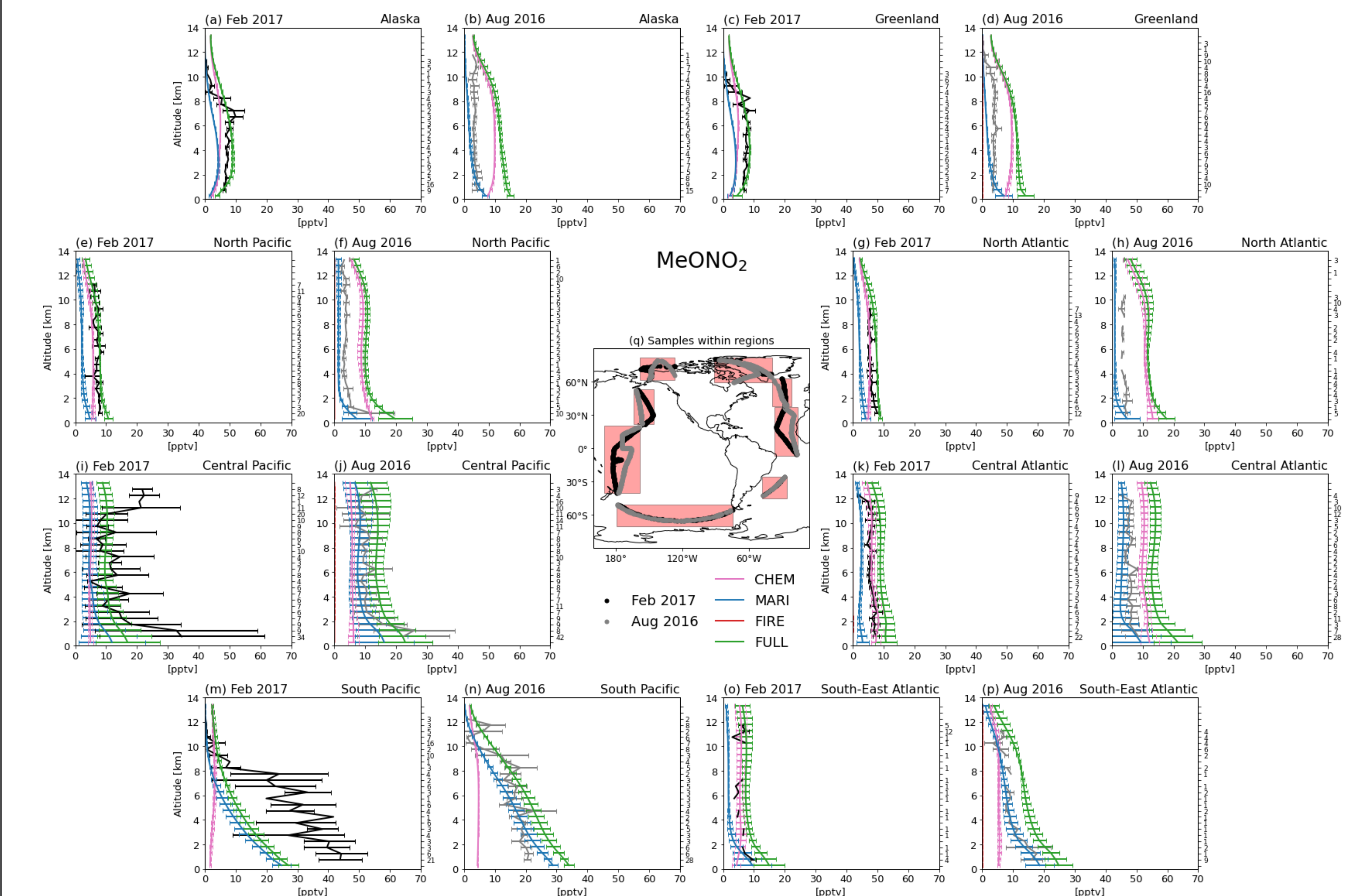
- add C₂-C₃ RONO₂ photochemistry and dry deposition to the chemical mechanism of the United Kingdom Chemistry and Aerosols (UKCA) model;
- derive C₁-C₃ RONO₂ oceanic emissions from Fisher et al. (2018) and biomass burning emissions from GFED4s (van der Werf et al. (2017));
- validate UKCA against NASA ATom-2;
- evaluate the impact of RONO₂ on O₃.

C₁-C₃ RONO₂ emissions



- Methyl nitrate (MeONO₂) oceanic and biomass burning emissions are stronger than emissions of other RONO₂ considered here.

UKCA vs ATom-2

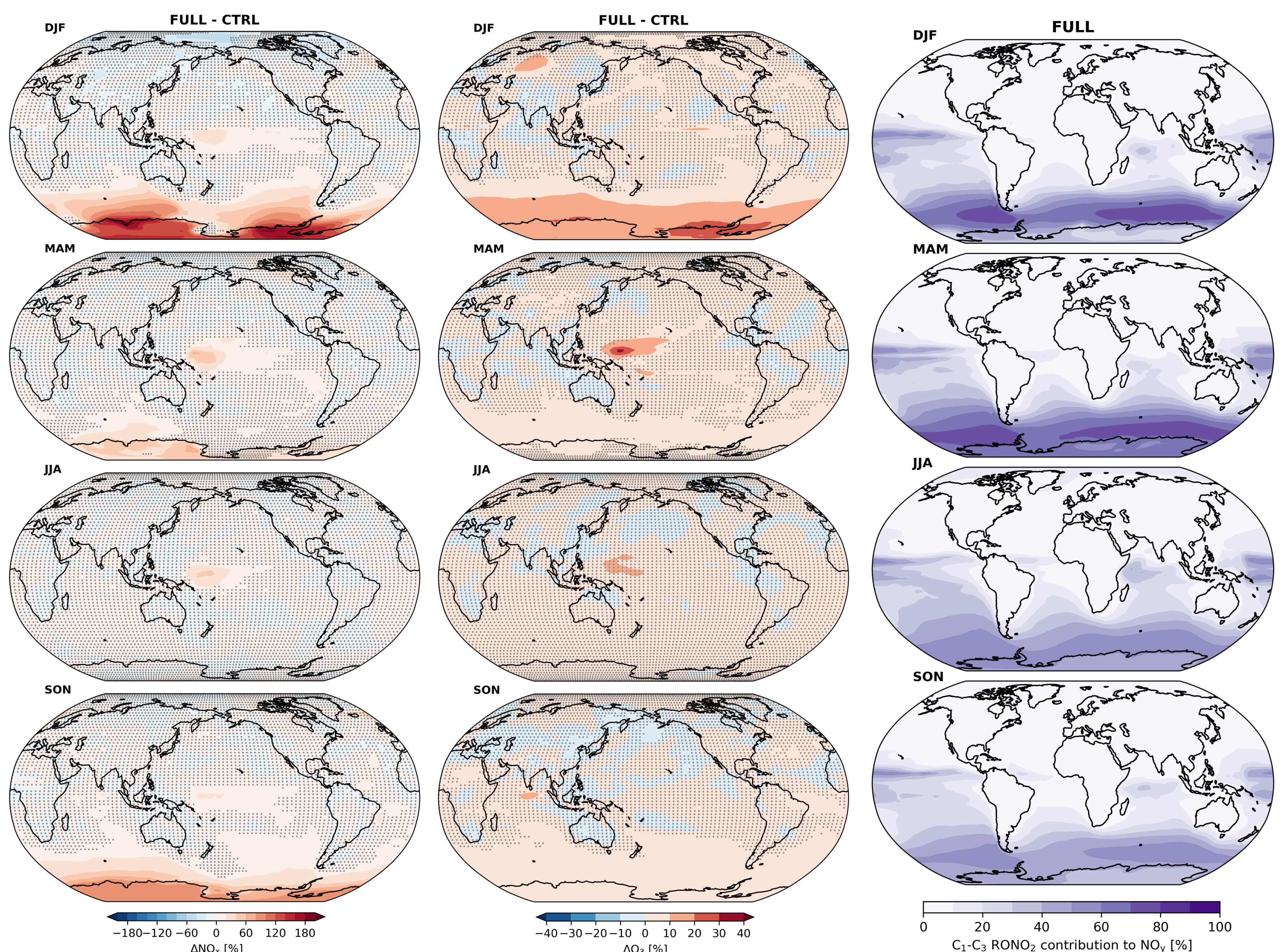


Local and global impacts of C₁-C₃ RONO₂ photochemistry and emissions

When C₁-C₃ RONO₂ photochemistry, dry deposition and oceanic and biomass burning emissions are included into the model (FULL simulation):

- NO_x and O₃ show a concurrent statistically significant* increase in the boundary layer (0-2 km) over the central Pacific and the Southern Ocean.
 - Over the central Pacific Ocean in MAM, NO_x increases by up to 260 ppt (66%) and O₃ increases by up to 3 ppb (32%).
 - Over the Southern Ocean in DJF, MAM and SON, with a maximum relative to other seasons increase in NO_x and O₃ occurring in DJF, with NO_x increasing by up to 7 ppt (189%) and O₃ increasing by up to 2 ppb (27%).
- C₁-C₃ RONO₂ contribute up to 78% to NO_y over the Southern Ocean.
- Annual mean tropospheric O₃ burden increases by 1.09±0.25%, while CH₄ lifetime decreases by 1.56±0.37%.

*Stippling shows areas where the difference between simulations with RONO₂ minus without RONO₂ is not statistically significant.



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